

## **Excerpts from:**

***Misconceptions About the Causes of Cancer.* LS Gold, TH Slone, NB Manley, and BN Ames. Vancouver, Canada: Fraser Institute (2002).**

### **Summary**

The major avoidable causes of cancer are: 1) smoking, which accounts for 27% of cancer deaths in Canada and 80-90% of lung cancer deaths; 2) dietary imbalances, which account for about another third, e.g., lack of sufficient amounts of dietary fruits and vegetables. 3) chronic infections, mostly in developing countries; and 4) hormonal factors, which are influenced primarily by life-style. This list may surprise readers who have come to think that synthetic chemicals like pesticide residues and water pollutants are major causes.

There is no cancer epidemic except for cancer of the lung due to smoking. (Cancer is actually many diseases, and the causes differ for cancers at different target sites.) Since 1971, overall cancer mortality rates in Canada (excluding lung cancer) have declined 17% in women and 5% in men. Regulatory policy that focuses on traces of synthetic chemicals is based on misconceptions about animal cancer tests. Current research indicates that it is not rare for substances to cause cancer in laboratory rodents in the standard high-dose experiments. Half of all chemicals tested, whether occurring naturally or produced synthetically, are “carcinogens”; there are high-dose effects in rodent cancer tests that are not relevant to low-dose human exposures and which may contribute to the high proportion of chemicals that test positive.

The focus of regulatory policy is on synthetic chemicals, but 99.9% of the chemicals humans ingest are natural. For example, more than 1000 naturally occurring chemicals have been described in coffee: 30 have been tested and 21 have been found to be carcinogenic in rodents in high-dose tests. Plants in the human diet contain thousands of natural “pesticides” produced by plants to protect themselves from insects and other predators: 72 have been tested and 38 have been found to give cancer to rodents. Thus, exposure to synthetic rodent carcinogens is small compared to the natural background of rodent carcinogens. High dose rodent cancer tests need to be re-evaluated by viewing results from this perspective.

In this study we highlight misconceptions about pollution, pesticides, and the causes of cancer. We briefly present the scientific evidence that undermines each misconception.

**Misconception 1—Cancer rates are soaring in the U.S. and Canada**

**Misconception 2—Synthetic chemicals at environmental exposure levels are an important cause of human cancer**

**Misconception 3—Reducing pesticide residues is an effective way to prevent diet-related cancer**

**Misconception 4—Human exposures to potential cancer hazards are primarily to synthetic chemicals**

**Misconception 5—The toxicology of synthetic chemicals is different from that of natural chemicals**

**Misconception 6—Cancer risks to humans can be assessed by standard high-dose animal cancer tests**

**Misconception 7—Synthetic chemicals pose greater carcinogenic hazards than natural chemicals**

## **Misconception 1—Cancer rates are soaring in the U.S. and Canada**

Overall cancer death rates in Canada (excluding lung cancer due to smoking) have declined 17% in women and 5% in men since 1971 (National Cancer Institute of Canada, 2001). In the United States the decline is similar: overall cancer death rates (excluding lung cancer) have declined 19% since 1950 (Ries *et al.*, 2000)

In Canada the types of cancer deaths that have decreased since 1971 are primarily stomach, cervical, and colorectal (National Cancer Institute of Canada, 2001). Those that have increased are primarily lung cancer (80-90% is due to smoking in Canada (American Cancer Society, 2000; Manuel and Hockin, 2000)), melanoma (probably due to sunburns), and non-Hodgkin's lymphoma (National Cancer Institute of Canada, 2001). If lung cancer is included, current cancer mortality rates (Ries *et al.*, 2000) are similar to those in 1972 (National Cancer Institute of Canada, 2001). For some cancers, mortality rates have begun to decline due in part to early detection, treatment and improved survival (American Cancer Society, 2000; Linet *et al.*, 1999), e.g., breast cancer in women (National Cancer Institute of Canada, 2001; Peto *et al.*, 2000). The rise in incidence rates in older age groups for some cancers, can be explained by known factors such as improved screening (Bailar and Gornik, 1997; Devesa *et al.*, 1995; Doll and Peto, 1981; Peto *et al.*, 2000). "The reason for not focusing on the reported incidence of cancer is that the scope and precision of diagnostic information, practices in screening and early detection, and criteria for reporting cancer have changed so much over time that trends in incidence are not reliable" (Bailar and Gornik, 1997, pp. 1569-70) Life expectancy has continued to rise since 1921 (Anderson, 1999; Manuel and Hockin, 2000): in Canada, life expectancy in the early 1920s was 59 years (<http://www.statcan.ca/english/Pgdb/People/Health/health26.htm>); today it is about 79 years (World Health Organization, 1984). Trends in the United States are similar to those in Canada (Anderson, 1999).

## **Misconception 2—Synthetic chemicals at environmental exposure levels are an important cause of human cancer**

Studies of cancer rates around the world indicate that the major avoidable causes of cancer primarily reflect lifestyle or other environmental factors that can be modified to reduce cancer risk (i.e. factors which are not genetic) (Armstrong and Doll, 1975; Doll and Peto, 1981). The main evidence for this conclusion is that rates of cancer in specific organs differ markedly in different countries; when people migrate to other countries their cancer rates change and within a few generations usually resemble the rates in their new countries. Additionally, rates change over time in a given country.

Neither epidemiology nor toxicology supports the idea that exposures to synthetic industrial chemicals at the levels at which they are generally found in the environment are important as a cause of human cancer (Ames *et al.*, 1995; Devesa *et al.*, 1995; Gold *et al.*, 1992).

Instead, other environmental factors have been identified in epidemiological studies that are likely to have a major effect on lowering cancer rates: reduction of smoking, improving diet (e.g., increased consumption of fruits and vegetables), hormonal factors (some of which are diet-related), and control of infections (Ames *et al.*, 1995). Few epidemiological studies find an association between cancer risk and low levels of industrial pollutants or pesticide residues; the associations are usually weak, the results are often conflicting, and the studies usually do not address

individual pesticides (Dich *et al.*, 1997). Moreover, the studies often do not correct for potentially large confounding factors such as composition of the diet (Ames, 1998; Ames *et al.*, 1995; Doll and Peto, 1981; Gold *et al.*, 2001a, <http://monographs.iarc.fr/monoeval/crthgr01.html>; International Agency for Research on Cancer, 1971-2001). Epidemiological studies on breast cancer risk have found no association with pesticide residues (Gammon *et al.*, 2002; Grodstein *et al.*, 1997; Hunter *et al.*, 1998). The most recent case-control study measured residues in blood of DDT, DDE, dieldrin, and chlordane and found no association with breast cancer (Gammon *et al.*, 2002).

From the toxicological perspective, exposures to synthetic pollutants are at very low levels and therefore rarely seem plausible as a causal factor, particularly when compared to the background of natural chemicals in the diet that are rodent carcinogens (Ames *et al.*, 1990a; Gold *et al.*, 1997b; Gold *et al.*, 1992). Even if one assumes that the worst-case risk estimates for synthetic pollutants are true risks, the proportion of cancer that the United States Environmental Protection Agency could prevent by regulation would be tiny (Gough, 1990). Historically, some high occupational exposures to some industrial chemicals have caused human cancer, though estimating the proportion of all cancers that are due to occupational exposures has been a controversial issue: a few percent seems a reasonable estimate (Ames *et al.*, 1995; Doll and Peto, 1981), and much of this is from asbestos in smokers. Exposures to synthetic chemicals or industrial mixtures in the workplace can be much higher than the exposure to chemicals in food, air, or water. Past occupational exposures have sometimes been high, and about half the agents that have been evaluated as human carcinogens by International Agency for Research on Cancer (IARC) were identified by workplace exposures. Since occupational cancer is concentrated among small groups with high levels of exposure, there is an opportunity to control or eliminate risks once they are identified. In the United States, Permissible Exposure Limits in the workplace are sometimes close to the carcinogenic dose in rodents (Gold *et al.*, 1994a), and thus require priority attention. See Misconception 7.

### **Aging and cancer**

Cancer is due, in part, to normal aging and increases exponentially with age in both rodents and humans (Ames *et al.*, 1993b). To the extent that the major avoidable risk factors for cancer are diminished, cancer will occur at later ages, and the proportion of cancer caused by normal metabolic processes will increase. Aging and its degenerative diseases appear to be due in part to oxidative damage to DNA and other macromolecules (Ames *et al.*, 1993b; Beckman and Ames, 1998). By-products of normal metabolism — superoxide, hydrogen peroxide, and hydroxyl radical — are the same oxidative mutagens produced by radiation. Mitochondria from old animals leak oxidants (Hagen *et al.*, 1997): old rats have been estimated to have about 66,000 oxidative DNA lesions per cell (Helbock *et al.*, 1998), although methods to measure such lesions are improving and may change the number somewhat. DNA is oxidized in normal metabolism because antioxidant defenses, though numerous, are not perfect. Antioxidant defenses against oxidative damage include vitamin C (Rice-Evans *et al.*, 1997) which comes from dietary fruits and vegetables, and vitamin E (Rice-Evans *et al.*, 1997), which comes from nuts, vegetable oils and fat. In addition, mitochondria, the organelles in the cell that generate energy and are the main source of oxidants, may need different antioxidants (Hagen *et al.*, 2002; Liu *et al.*, 2002a; Liu *et al.*, 2002b). Increasing antioxidant intake in those persons with low intakes may help to prevent cancer, but it is difficult to disentangle dietary intake of individual vitamins or minerals in epidemiological studies (Ames and Wakimoto, 2002).

## Smoking

In Canada smoking contributes to 27% of cancer deaths and about 45,000 premature deaths per year (American Cancer Society, 2000; Makomaski Illing and Kaiserman, 1999; National Cancer Institute of Canada, 2000; Ries *et al.*, 2000). Overall, 21% of deaths from the 3 leading causes of death (cancer, heart disease and cerebrovascular disease) are attributable to smoking (Makomaski Illing and Kaiserman, 1999). Tobacco is a cause of cancer of the lung, mouth, pharynx, larynx, esophagus, bladder, pancreas, stomach, kidney, uterine cervix, and myeloid leukemia (International Agency for Research on Cancer, 1986; International Agency for Research on Cancer, 2002, in press). Smoke contains a wide variety of mutagens and substances that are carcinogenic in rodents. Smoking is also a severe *oxidative stress* and causes inflammation in the lung. The oxidants in cigarette smoke—mainly nitrogen oxides—deplete the body's antioxidants (Lykkesfeldt *et al.*, 2000). Thus, smokers need to ingest more vitamin C than non-smokers to achieve the same level in blood, but they do not tend to do so. An inadequate concentration of vitamin C in plasma is more common among smokers (Lykkesfeldt *et al.*, 2000). A recent Danish study indicated that smokers consumed fewer fruits and vegetables than nonsmokers (Osler *et al.*, 2002). Additionally, people who take supplements of vitamins and minerals are less likely to be smokers (Patterson *et al.*, 2001). Men with inadequate diets or who smoke may damage the DNA in all cells of the body, including their sperm. When the level of dietary vitamin C is insufficient to keep seminal fluid vitamin C at an adequate level, the oxidative lesions in sperm DNA are increased 2.5 times (Ames *et al.*, 1994; Fraga *et al.*, 1991; Fraga *et al.*, 1996). Male smokers have more oxidative lesions in sperm DNA (Fraga *et al.*, 1996) and more chromosomal abnormalities in sperm (Wyrobek *et al.*, 1995) than do nonsmokers. It is plausible, therefore, that fathers who smoke may increase the risk of birth defects and childhood cancer in offspring (Ames *et al.*, 1994; Fraga *et al.*, 1991; Woodall and Ames, 1997). Some epidemiological studies suggest that the rate of childhood cancers is increased in offspring of male smokers (Ji *et al.*, 1997; Sorahan *et al.*, 1995).

Involuntary (environmental) exposure to tobacco smoke has also been evaluated as a human carcinogen (International Agency for Research on Cancer, 2002, in press; U.S. Department of Health and Human Services, 1986; U.S. Environmental Protection Agency, 1992b), and is estimated to increase the risk of lung cancer by 20-30%. In comparison, smokers have an increased risk of lung cancer of 2000% (International Agency for Research on Cancer, 2002, in press), i.e. 600 to 1000 times greater risk than from involuntary smoking.

## Diet

Dietary factors have been estimated to account for about one-third of cancer deaths in the United States (American Cancer Society, 2000; Ames *et al.*, 1995; Doll and Peto, 1981; Ries *et al.*, 2000), and specific dietary factors are slowly being clarified, although epidemiological research on diet has many complexities and confounding factors. Low intake of fruits and vegetables is associated with increased cancer incidence in many case-control studies (Block *et al.*, 1992; World Cancer Research Fund, 1997); results from several recent cohort studies, however, have been less consistent (Willett, 2001). (See Misconception 3). Excessive consumption of alcoholic beverages is associated with cancers of the breast, oral cavity (primarily in smokers), and liver (International Agency for Research on Cancer, 1988; Willett, 2001).

There has been considerable interest in calories (and dietary fat) as a risk factor for cancer, in part because caloric restriction markedly lowers the cancer rate and increases life span in rodents (Ames *et al.*, 1995; Hart *et al.*, 1995b; Turturro *et al.*, 1996; Vainio and Bianchini, 2002).

For two common cancers, breast and colon, international comparisons in incidence suggested a role for fat intake; however, combined analyses of many studies do not support such an association (Hunter *et al.*, 1996; Willett, 2001). Higher intake of dietary fiber does not appear to protect against colon cancer, although some earlier case-control studies suggested that it did (Willett, 2001). Current scientific attention has focused on body weight (obesity), adult weight gain, and inadequate physical activity as risk factors for cancer (Caan *et al.*, 1998; Giovannucci *et al.*, 1995; Huang *et al.*, 1997; Vainio and Bianchini, 2002; Willett, 2001). A recent report by IARC states: "Taken together, excess body weight and physical inactivity account for approximately one fourth to one third of breast cancer, cancers of the colon, endometrium, kidney (renal cell) and oesophagus (adenocarcinoma). Thus adiposity and inactivity appear to be the most important avoidable causes of postmenopausal breast cancer, endometrial cancer, renal cell cancer, and adenocarcinoma of the oesophagus, and among the most important avoidable causes of colon cancer." (Vainio and Bianchini, 2002) Lack of regular physical activity contributes independently to risk of colon (Giovannucci *et al.*, 1995; Giovannucci *et al.*, 1996; Martinez *et al.*, 1997; Platz *et al.*, 2000; Willett, 2001) and breast cancer (Bernstein *et al.*, 1994; Rockhill *et al.*, 1999; Willett, 2001).

### **Hormonal factors**

Endogenous reproductive hormones play a large role in cancer, including that of the breast, prostate, ovary, and endometrium (Henderson and Feigelson, 2000; Henderson *et al.*, 1991), contributing to about 20% of all cancer. Many lifestyle factors such as reproductive history, lack of exercise, obesity, and alcohol influence hormone levels and therefore affect risk (Ames *et al.*, 1995; Henderson and Feigelson, 2000; Henderson *et al.*, 1991; Hunter and Willett, 1993; Kelsey and Bernstein, 1996; Writing Group for the Women's Health Initiative Investigators, 2002). The mechanisms for postmenopausal breast cancer may involve changes in hormone metabolism, e.g. earlier menstruation and postmenopausal release of estrogen from body fat, never having a child or first birth over age 35. Recent results of a clinical trial in the Women's Health Initiative study indicate that hormone replacement therapy (estrogen and progestin) increases the risk of postmenopausal breast cancer (Writing Group for the Women's Health Initiative Investigators, 2002).

### **Chronic inflammation**

Chronic inflammation results in the release of oxidative mutagens from white cells and other sentinel cells of the immune system, which combat bacteria, parasites, and viruses by destroying them with potent, mutagenic oxidizing agents (Ames *et al.*, 1995; Christen *et al.*, 1999). These oxidants protect humans from immediate death from infection, but they also cause oxidative damage to DNA, chronic cell killing with compensatory cell division, and mutation (Shacter *et al.*, 1988; Yamashina *et al.*, 1986); thus they contribute to cancer. Anti-inflammatory agents, including some antioxidants, appear to inhibit some of the pathology of chronic inflammation. Chronic infections that give rise to chronic inflammation are estimated to cause about 21% of new cancer cases in developing countries and 9% in developed countries (Pisani *et al.*, 1997), e.g. hepatitis B and C, viruses and liver cancer, *Helicobacter pylori* and stomach cancer. Obesity is associated with a systemic chronic inflammation, which suggests that it may play a role in cancer risk (Das, 2001).

### **Other factors**

Other causal factors in human cancer are excessive sun exposure, viruses (e.g., human papillomavirus and cervical cancer), and pharmaceuticals (e.g. phenacetin, some chemotherapy agents, diethylstilbestrol, estrogens). Genetic factors affect susceptibility to cancer and interact with lifestyle and other risk factors. Biomedical research is uncovering important genetic variation in humans that can affect susceptibility.

### **Misconception 3: Reducing pesticide residues is an effective way to prevent diet-related cancer.**

Reduction in the use of pesticides will not effectively prevent diet-related cancer. Diets high in fruits and vegetables, which are the source of most pesticide residue exposures to humans, are associated with reduced risk of many types of cancer. Less use of synthetic pesticides would increase costs of fruits and vegetables and thus likely reduce consumption, especially among people with low incomes, who spend a higher percentage of their income on food.

### **Dietary fruits and vegetables and cancer prevention**

Two types of evidence, (1) epidemiological studies on diet and cancer and (2) laboratory studies on vitamin or mineral inadequacy, support the idea that low intake of fruits and vegetables is associated with increased risk of degenerative diseases, including cancer, cardiovascular disease, cataracts, and brain dysfunction (Ames *et al.*, 1995; Ames *et al.*, 1993b; Ames and Wakimoto, 2002). Fruits and vegetables are an important source of essential vitamins and minerals (Ames and Wakimoto, 2002).

Despite the evidence about the importance of fruits and vegetables, the Canadian campaign “5-to-10-a-Day: Are You Getting Enough?” reported that 67% of Canadians do not eat 5 or more servings of fruits and vegetables per day, based on a Nielson telephone survey of women ([http://5to10aday.com/eng/media\\_news\\_nr1.htm](http://5to10aday.com/eng/media_news_nr1.htm); A. Matyas, pers.comm.). Another survey, by interview, reported that about half of Canadians do not eat 5 or more per day (Gray-Donald *et al.*, 2000). In the United States, it has been estimated that eighty percent of children and adolescents, and 68% of adults (Krebs-Smith *et al.*, 1995; Krebs-Smith *et al.*, 1996) do not eat 5 or more per day. Publicity about hundreds of minor, hypothetical risks, such as pesticide residues, can result in a loss of perspective on what is important: (U.S. National Cancer Institute, 1996) only 7% of Canadians surveyed thought that eating fruits and vegetables can reduce the risk of cancer ([http://www.5to10aday.com/eng/media\\_executive\\_summary.htm](http://www.5to10aday.com/eng/media_executive_summary.htm)). There is a paradox in the public concern about possible cancer hazards from the low levels of pesticide residues in food and the lack of public understanding of the evidence of cancer protection by eating *more* of the main foods that contain pesticide residues — fruits and vegetables.

Several reviews of the epidemiological literature show that a high proportion of case-control studies find an inverse association between fruit and vegetable consumption and cancer risk (Block *et al.*, 1992; Hill *et al.*, 1994; Steinmetz and Potter, 1996; World Cancer Research Fund, 1997). It is not clear from these studies whether individuals who consume very low amounts are the only people at risk, i.e. whether there is an adequate level above which there is no increased cancer risk. A recent international panel considered the evidence of a protective effect of fruits and vegetables most convincing for cancers of the oral cavity, esophagus, stomach, and lung (World Cancer Research Fund, 1997). In another review, the median relative risk was about 2 for the quarter of the population with the lowest dietary intake of fruits and

vegetables vs. the quarter with the highest intake for cancers of the lung, larynx, oral cavity, esophagus, stomach, bladder, pancreas, and cervix (Block et al., 1992). The median relative risk was not as high for the hormonally-related cancers of breast, prostate, and ovary, or for the colon.

More than 30 large cohort studies of the relationship between diet and cancer are in progress in various countries (Willett, 2001). Generally the results of cohort studies have been less strong and less consistent in their findings about the association between fruit and vegetable intake and cancer risk than case-control studies, (Botterweck et al., 1998; Galanis et al., 1998; Giovannucci et al., 2002; Jansen et al., 2001; Kasum et al., 2002; McCullough et al., 2001; Michels et al., 2000; Ozasa et al., 2001; Schuurman et al., 1998; Sellers et al., 1998; Smith-Warner et al., 2001; Terry et al., 1998; Terry et al., 2001; Voorrips et al., 2000; Zeegers et al., 2001). Some cohort studies have shown a lack of association between fruit and vegetable consumption and cancers of the colon, breast, and stomach (Botterweck et al., 1998; Galanis et al., 1998; Kasum et al., 2002; McCullough et al., 2001; Michels et al., 2000; Sellers et al., 1998; Smith-Warner et al., 2001; Terry et al., 1998; Terry et al., 2001; Voorrips et al., 2000). As more analyses are reported from cohort studies, the estimation of relative risks should become more precise.

Observational epidemiological studies have many limitations that make interpretation of results complex. Unlike experiments in rodents, in which a single variable is changed and everything else is controlled for, in epidemiological studies on diet, people eat varied diets and change over time, they may not recall correctly their eating habits, and they have different genetic makeups. Some examples of the kinds of complexities in these studies follow. The category “fruits and vegetables” is a broad one and if inadequacy of vitamin and mineral intake is actually the underlying protective factor in fruits and vegetables, then it is the individual foods rather than the combined intake that may be related to a specific cancer (Willett, 2001). This is usually not the focus in research investigations; rather, the focus is the combined category, fruits and vegetables. Additionally, use of a multivitamin pill or of a particular vitamin pill, has generally not been taken into account in these studies, and this may confound the results because supplement users have a healthier lifestyle, including greater intake of fruits and vegetables as well as other factors like lower rates of smoking, lower fat diets, and a belief in the connection between diet and cancer that may affect both their behaviors and their recall of dietary intakes (Block et al., 1994; Patterson et al., 2001). Methodological limitations of case-control studies that may account for the stronger findings than in cohort studies include recall bias (controls may remember their dietary habits differently from cases — the people with cancer) and selection bias (people who choose to participate as controls may have healthier lifestyles that include among other factors, higher intake of fruits and vegetables, thus leading to a lower observed relative risk that may not really be due to fruits and vegetables).

### **Vitamin and mineral inadequacy and cancer risk**

Laboratory studies of vitamin and mineral inadequacy indicate an association with DNA damage, which suggests that the vitamin and mineral content of fruits and vegetables may underlie the observed association between fruit and vegetable intake and cancer risk. Maximum health and lifespan require metabolic harmony, and inadequate or sub-optimal intake of essential vitamins and minerals may result in metabolic damage that can affect many functions and hence affect disease development.

Antioxidants such as vitamin C (whose dietary source is fruits and vegetables), vitamin E, and selenium protect against oxidative damage caused by normal metabolism (Helbock *et al.*, 1998), smoking (Ames, 1998), and inflammation (Ames *et al.*, 1993b) (See Misconception #2). Deficiency of some vitamins and minerals can mimic radiation in damaging DNA by causing single- and double-strand breaks, or oxidative lesions, or both (Ames, 1998). Those vitamins and minerals whose deficiency appears to mimic radiation are folic acid, B12, B6, niacin, C, E, iron, and zinc, with the laboratory evidence ranging from likely to compelling. In the United States, the percentage of the population that consumes less than half the RDA in the diet (i.e. ignoring supplement use) for five of these eight vitamins or minerals is estimated to be: zinc (10% of women/men 50+ years old), iron (25% of women 20-30 years, and 5% of women 50+), vitamin C (25% of women/men 20+ years), folate (50% of women 20+ years; 25% of men 20+ years), vitamin B6 (10% of women/men 20+), vitamin B12 (10% of women 20+; 5% of men 20+) (Ames and Wakimoto, 2002). These deficiencies may constitute a considerable percentage of the United States population (Ames, 1998; Ames and Wakimoto, 2002).

Folic acid deficiency, one of the most common vitamin deficiencies in the population consuming few dietary fruits and vegetables, causes chromosome breaks in humans (Blount *et al.*, 1997). The mechanism of chromosome breaks has been shown to be analogous to radiation (Blount *et al.*, 1997). Folate supplementation above the RDA minimized chromosome breakage (Fenech *et al.*, 1998). Folate deficiency has been associated with increased risk of colon cancer (Giovannucci *et al.*, 1993; Mason, 1994): in the Nurses' Health Study women who took a multi-vitamin supplement containing folate for 15 years had a 75% lower risk of colon cancer (Giovannucci *et al.*, 1998). Folate deficiency also damages human sperm (Wallock *et al.*, 2001), causes neural tube defects in the fetus and an estimated 10% of United States heart disease (Boushey *et al.*, 1995). Approximately 10% of the U.S. population (Senti and Pilch, 1985) had a lower folate level than that at which chromosome breaks occur (Blount *et al.*, 1997). Nearly 20 years ago, two small studies of low-income (mainly African-American) elderly (Bailey *et al.*, 1979) and adolescents (Bailey *et al.*, 1982) showed that about half the people in both groups studied had folate levels that low. Recently in Canada and the U.S., flour, rice, pasta, and corn-meal have been supplemented with folate (Health Canada, 1998; Jacques *et al.*, 1999).

Recent evidence indicates that vitamin B6 deficiency works by the same mechanism as folate deficiency and this would cause chromosome breaks (Mashiyama, Shultz & Ames, unpublished). Niacin contributes to the repair of DNA strand-breaks by maintaining nicotinamide adenine dinucleotide levels for the poly ADP-ribose protective response to DNA damage (Zhang *et al.*, 1993). As a result, dietary insufficiencies of niacin (15% of some populations are deficient) (Jacobson, 1993), folate, and antioxidants may interact synergistically to adversely affect DNA synthesis and repair. Diets deficient in fruits and vegetables are commonly low in folate, antioxidants, (e.g., vitamin C), and many other vitamins and minerals, result in DNA damage, and are associated with higher cancer rates (Ames, 1998; Ames *et al.*, 1995; Block *et al.*, 1992; Subar *et al.*, 1989).

### **Vitamins and Minerals from dietary sources other than fruits and vegetables**

Vitamins and minerals whose main dietary sources are other than fruits and vegetables, are also likely to play a significant role in the prevention and repair of DNA damage, and thus are important to the maintenance of long-term health (Ames, 1998). Deficiency of vitamin B12 causes a functional folate deficiency, accumulation of homocysteine (a risk factor for heart disease) (Herbert and Filer, 1996), and chromosome breaks. B12 supplementation above the RDA was



necessary to minimize chromosome breakage (Fenech *et al.*, 1998). Strict vegetarians are at increased risk for developing vitamin B12 deficiency since the dietary source is animal products (Herbert and Filer, 1996).

Epidemiological studies of supplement usage (vitamin and mineral intake by pill) have shown at most only modest support for an association. The strongest protective effect was for vitamin E and cancers of the prostate and colon (Patterson *et al.*, 2001). There are many potential problems in conducting such studies including the need and difficulty in measuring supplement use over a long period of time, potential confounding of supplement usage with many other aspects of a healthy lifestyle which are related to it, such as more exercise, better diet, and not smoking (Patterson *et al.*, 2001). Clinical trials of supplements are generally too short to measure cancer risk since cancers usually develop slowly and the risk increases with age; moreover, such trials cannot measure the potential reduction in risk if supplements are taken throughout a lifetime (Block, 1995). Additionally, cancer risks of supplement users may be overestimated because they are more likely to undergo early screening like mammograms or tests for prostate cancer (prostate-specific antigen, PSA) which are associated with increased diagnosis (Patterson *et al.*, 2001). Such confounding factors are not measured in many epidemiological studies.

Intake of adequate amounts of vitamins and minerals may have a major effect on health, and the costs and risks of a daily multivitamin/mineral pill are low (Ames, 1998). More research in this area, as well as efforts to improve diets, should be high priorities for public policy.

#### **Misconception 4: Human exposures to potential cancer hazards are primarily to synthetic chemicals.**

Contrary to common perception, 99.9% of the chemicals humans ingest are natural. The amounts of synthetic pesticide residues in plant foods, for example, are extremely low compared to the amounts of natural “pesticides” produced by plants themselves (Ames *et al.*, 1990a; Ames *et al.*, 1990b; Gold *et al.*, 1999; Gold *et al.*, 1997b; Gold and Zeiger, 1997). Of all dietary pesticides that humans eat, 99.99% are natural: these are chemicals produced by plants to defend themselves against fungi, insects, and other animal predators (Ames *et al.*, 1990a; Ames *et al.*, 1990b). Each plant produces a different array of such chemicals. On average, the Western diet includes roughly 5,000 to 10,000 different natural pesticides and their breakdown products. Americans eat about 1,500 mg of natural pesticides per person per day, which is about 10,000 times more than they consume of synthetic pesticide residues (Ames *et al.*, 1990b). Even though only a small proportion of natural pesticides has been tested for carcinogenicity, half of those tested (38/72) are rodent carcinogens; naturally occurring pesticides that are rodent carcinogens are ubiquitous in fruits, vegetables, herbs, and spices (Gold *et al.*, 1997b; Gold *et al.*, 1992) (Table 2).

Cooking of foods produces burnt material (about 2,000 mg per person per day) that contains many rodent carcinogens.

In contrast, the residues of 200 synthetic chemicals measured by United States Federal Drug Administration, including the synthetic pesticides thought to be of greatest importance, average only about 0.09 mg per person per day (Ames *et al.*, 1990a; Gold *et al.*, 1997b; Gold *et al.*, 1992). In a single cup of coffee, the natural chemicals that are rodent carcinogens are about equal in weight to an entire year’s worth of synthetic pesticide residues that are rodent carcinogens, even though only 3% of the natural chemicals in roasted coffee have been adequately

tested for carcinogenicity (Gold *et al.*, 1992) (Table 3). This does not mean that coffee or natural pesticides are a cancer risk for humans, but rather that assumptions about high-dose animal cancer tests for assessing human risk at low doses need reexamination. No diet can be free of natural chemicals that are rodent carcinogens (Gold *et al.*, 1999; Gold *et al.*, 1997b; Gold and Zeiger, 1997).

The emphasis in cancer bioassays of testing synthetic chemicals means that only minimal data are available on the enormous background of naturally occurring chemicals. If many of the natural chemicals were tested, it is likely that many dietary constituents would be carcinogens in high-dose animal tests. The importance for human cancer of any single rodent carcinogen in the diet is questionable because of the ubiquitous occurrence of so many naturally occurring chemicals that have not been tested and the fact that half of those tested are positive in such tests (Misconception 6).

**Table 2. Carcinogenicity status of natural pesticides tested in rodents <sup>a</sup>**

<b>Carcinogens:</b> <b>N=38</b>	acetaldehyde methylformylhydrazone, allyl isothiocyanate, arecoline.HCl, benzaldehyde, benzyl acetate, caffeic acid, capsaicin, catechol, clivorine, coumarin, crotonaldehyde, 3,4-dihydrocoumarin, estragole, ethyl acrylate, <i>N</i> 2- $\gamma$ -glutamyl- <i>p</i> -hydrazinobenzoic acid, hexanal methylformylhydrazine, <i>p</i> -hydrazinobenzoic acid.HCl, hydroquinone, 1-hydroxyanthraquinone, lasiocarpine, <i>d</i> -limonene, 3-methoxycatechol, 8-methoxypsoralen, <i>N</i> -methyl- <i>N</i> -formylhydrazine, $\alpha$ -methylbenzyl alcohol, 3-methylbutanal methylformylhydrazone, 4-methylcatechol, methyl eugenol, methylhydrazine, monocrotaline, pentanal methylformylhydrazone, petasitenine, quercetin, reserpine, safrole, senkirkine, sesamol, symphytine
<b>Noncarcinogens:</b> <b>N=34</b>	atropine, benzyl alcohol, benzyl isothiocyanate, benzyl thiocyanate, biphenyl, <i>d</i> -carvone, codeine, deserpidine, disodium glycyrrhizinate, ephedrine sulphate, epigallocatechin, eucalyptol, eugenol, gallic acid, geranyl acetate, $\beta$ - <i>N</i> -[ $\gamma$ - <i>l</i> (+)-glutamyl]-4-hydroxymethylphenylhydrazine, glycyrrhetic acid, <i>p</i> -hydrazinobenzoic acid, isosafrole, kaempferol, <i>dl</i> -menthol, nicotine, norharman, phenethyl isothiocyanate, pilocarpine, piperidine, protocatechuic acid, rotenone, rutin sulfate, sodium benzoate, tannic acid, 1-trans- $\delta^9$ -tetrahydrocannabinol, turmeric oleoresin, vinblastine

***These rodent carcinogens occur in: absinthe, allspice, anise, apple, apricot, banana, basil, beet, broccoli, Brussels sprouts, cabbage, cantaloupe, caraway, cardamom, carrot, cauliflower, celery, cherries, chili pepper, chocolate, cinnamon, citronella, cloves, coffee, collard greens, comfrey herb tea, corn, coriander, currants, dill, eggplant, endive, fennel, garlic, grapefruit, grapes, guava, honey, honeydew melon, horseradish, kale, lemon, lentils, lettuce, licorice, lime, mace, mango, marjoram, mint, mushrooms, mustard, nutmeg, onion, orange, oregano, paprika, parsley, parsnip, peach, pear, peas, black pepper, pineapple, plum, potato, radish, raspberries, rhubarb, rosemary, rutabaga, sage, savory, sesame seeds, soybean, star anise, tarragon, tea, thyme, tomato, turmeric, and turnip.***

<sup>a</sup> Fungal toxins are not included. From the Carcinogenic Potency Database (Gold *et al.*, 1999; Gold and Zeiger, 1997).

**Table 3. Carcinogenicity in rodents of natural chemicals in roasted coffee <sup>a</sup>**

<b>Positive: N=21</b>	acetaldehyde, benzaldehyde, benzene, benzofuran, benzo( <i>a</i> )pyrene, caffeic acid, catechol, 1,2,5,6-dibenzanthracene, ethanol, ethylbenzene, formaldehyde, furan, furfural, hydrogen peroxide, hydroquinone, isoprene, limonene, 4-methylcatechol, styrene, toluene, xylene
<b>Not positive: N=8</b>	acrolein, biphenyl, choline, eugenol, nicotinamide, nicotinic acid, phenol, piperidine
<b>Uncertain:</b>	caffeine
<b>Yet to test:</b>	~ 1000 chemicals

<sup>a</sup> From the Carcinogenic Potency Database (Gold *et al.*, 1999; Gold and Zeiger, 1997).

### **Misconception 5: The toxicology of synthetic chemicals is different from that of natural chemicals.**

It is often assumed that because natural chemicals are part of human evolutionary history, whereas synthetic chemicals are recent, the mechanisms that have evolved in animals to cope with the toxicity of natural chemicals will fail to protect against synthetic chemicals (Ames *et al.*, 1987, and Letters). This assumption is flawed for several reasons (Ames *et al.*, 1996; Ames *et al.*, 1990b; Gold *et al.*, 1997b).

#### **Natural defenses are general rather than specific for each chemical**

Humans have many natural defenses that buffer against normal exposures to toxins (Ames *et al.*, 1990b); these usually are general rather than tailored to each specific chemical. Thus, the defenses work against both natural and synthetic chemicals. Examples of general defenses include the continuous shedding of cells exposed to toxins — the surface layers of the mouth, esophagus, stomach, intestine, colon, skin, and lungs are discarded every few days; DNA repair enzymes, which repair DNA that has been damaged from many different sources; and detoxification enzymes of the liver and other organs which generally target classes of toxins rather than individual toxins. That defenses are usually general, rather than specific for each chemical, makes good evolutionary sense. The reason that predators of plants evolved general defenses presumably was to be prepared to counter a diverse and ever-changing array of plant toxins in an evolving world; if a herbivore had defenses against only a set of specific toxins, it would be at a great disadvantage in obtaining new food when favored foods became scarce or evolved new toxins.

#### **Natural agents can be carcinogenic to humans**

Various natural agents that have been present throughout vertebrate evolutionary history nevertheless cause cancer in vertebrates (Ames *et al.*, 1990b; Gold *et al.*, 1999; Gold *et al.*, 1997a; Vainio *et al.*, 1995). Mold toxins, such as aflatoxin, have been shown to cause cancer in rodents and other species, including humans (Gold *et al.*, 1999) (See Table 4). Despite their presence throughout evolution, many of the common elements are carcinogenic to humans at high doses (e.g., salts of cadmium, beryllium, nickel, chromium, and arsenic). Furthermore, epidemiologi-

cal studies from various parts of the world show that certain natural chemicals in food may be carcinogenic risks to humans; for example, the chewing of betel nuts with tobacco is associated with oral cancer, and Chinese-style salted fish is associated with nasopharyngeal cancer (Gold *et al.*, 2001a, <http://monographs.iarc.fr/monoeval/crthgr01.html>).

Humans have not had time to evolve a “toxic harmony” with all of the plants in their diet. The human diet has changed markedly in the last few thousand years. Indeed, very few of the plants that humans eat today (e.g., coffee, cocoa, tea, potatoes, tomatoes, corn, avocados, mangoes, olives, and kiwi fruit), would have been present in a hunter-gatherer’s diet. Natural selection works far too slowly for humans to have evolved specific resistance to the food toxins in these relatively newly introduced plants.

Since no plot of land is free from attack by insects, plants need chemical defenses — either natural or synthetic — in order to survive. Thus, there is a trade-off between naturally occurring and synthetic pesticides. One consequence of disproportionate concern about synthetic pesticide residues is that some plant breeders develop plants to be more insect-resistant by making them higher in natural toxins. A case study illustrates the potential hazards of this approach to pest control: When a major grower introduced a new variety of highly insect-resistant celery into commerce, people who handled the celery developed rashes when they were subsequently exposed to sunlight. Some detective work found that the pest-resistant celery contained 6200 parts per billion (ppb) of carcinogenic (and mutagenic) psoralens instead of the 800 ppb present in common celery (Gold *et al.*, 1999; Gold *et al.*, 1997b).

**Table 4. Proportion of chemicals evaluated as carcinogenic <sup>a</sup>**

Chemicals tested in both rats and mice <sup>a</sup>	
Chemicals in the CPDB	<b>350/590 (59%)</b>
Naturally-occurring chemicals in the CPDB	<b>79/139 (57%)</b>
Synthetic chemicals in the CPDB	<b>271/451 (60%)</b>
Chemicals tested in rats and/or mice <sup>a</sup>	
Chemicals in the CPDB	<b>702/1348 (52%)</b>
Natural pesticides in the CPDB	<b>38/72 (53%)</b>
Mold toxins in the CPDB	<b>14/23 (61%)</b>
Chemicals in roasted coffee in the CPDB	<b>21/30 (70%)</b>
Commercial pesticides	<b>79/194 (41%)</b>
Innes negative chemicals retested <sup>a</sup>	<b>17/34 (50%)</b>
Physician’s Desk Reference (PDR): drugs with reported cancer tests <sup>b</sup>	<b>117/241 (49%)</b>
FDA database of drug submissions <sup>c</sup>	<b>125/282 (44%)</b>

<sup>a</sup> From the Carcinogenic Potency Database (Gold *et al.*, 1999; Gold and Zeiger, 1997).

<sup>b</sup> From Davies and Monro (Davies and Monro, 1995).

<sup>c</sup> From Contrera *et al.* (Contrera *et al.*, 1997).

140 drugs are in both the FDA and PDR databases.

### **Misconception 6: Cancer risks to humans can be assessed by standard high-dose animal cancer tests.**

Approximately half of all chemicals that have been tested in standard animal cancer tests, whether natural or synthetic, are rodent carcinogens (Table 4) (Gold *et al.*, 1989a; Gold *et al.*, 1999; Gold *et al.*, 1997a). Why such a high positivity rate? A reasonable explanation is that the design of these experiments produces effects that would not occur at lower doses. In standard cancer tests, rodents are given chronic, near-toxic doses, the maximum tolerated dose (MTD). The rationale for this experimental design was based on a consensus in the 1970s that chemicals with carcinogenic potential would be rare and therefore experiments had to be designed to maximize the chance of finding an effect. Since costs of conducting these tests are high (currently 2-4 million dollars per chemical) (U.S. National Toxicology Program, 1998), a limited number of animals would be put on test (50 per dose group, for controls, a high dose, and one-half the high dose). Because of the small number of animals on test, the studies lack statistical power, and therefore the doses were set as high as the animals would tolerate yet still live long enough to have time to get cancer, since cancer is a disease of old-age. Evidence is accumulating that cell division caused by the high dose itself, rather than the chemical *per se*, is increasing the carcinogenic effects, and therefore the positivity rate. High doses can cause chronic wounding of tissues, cell death, and consequent chronic cell division of neighboring cells. This is a risk factor for cancer (Ames *et al.*, 1996) because, each time a cell divides the probability increases that a mutation will occur, thereby increasing the risk for cancer.

At the low levels to which humans are usually exposed, such increased cell division does not occur. The process of mutagenesis and carcinogenesis is complicated because many factors are involved: e.g., DNA lesions, DNA repair, cell division, clonal instability, apoptosis (cell suicide in response to DNA damage), and p53 (a cell cycle control gene that is mutated in half of human tumors) (Christensen *et al.*, 1999; Hill *et al.*, 1999). The normal endogenous level of oxidative DNA lesions in cells is appreciable (Helbock *et al.*, 1998). In addition, tissues injured by high doses of chemicals have an inflammatory immune response involving activation of white cells in response to cell death (Adachi *et al.*, 1995; Czaja *et al.*, 1994; Gunawardhana *et al.*, 1993; Laskin and Pendino, 1995; Laskin *et al.*, 1988; Roberts and Kimber, 1999; Wei *et al.*, 1993a; Wei *et al.*, 1993b). Activated white cells release mutagenic oxidants (including peroxynitrite, hypochlorite, and H<sub>2</sub>O<sub>2</sub>). Therefore, the very low levels of chemicals to which humans are exposed through water pollution or synthetic pesticide residues may pose no or only minimal cancer risks because these effects do not occur at low doses.

Analyses of the limited data on dose-response in bioassays are consistent with the idea that cell division from cell-killing and cell replacement is important. Among rodent bioassays with two doses and a control group, about half the sites evaluated as target sites are statistically significant at the MTD but not at half the MTD ( $p < 0.05$ ). Ad libitum feeding in the standard bioassay can also contribute to the high positivity rate (Hart *et al.*, 1995a). In calorie-restricted mice, cell division rates are markedly lower in several tissues than in ad libitum-fed mice (Lok *et al.*, 1990). Linearity of response to increasing dosage seems less likely than has been assumed because of the inducibility of the numerous defense enzymes which deal with exogenous chemicals as groups, e.g., oxidants, electrophiles, and thus protect us against the natural world of mutagens as well as the small amounts of synthetic chemicals (Ames *et al.*, 1990b; Luckey, 1999; Munday and Munday, 1999; Trosko, 1998).

More than a decade ago, we argued that risk assessment for humans requires data on mechanism of carcinogenesis for each chemical (Ames and Gold, 1990; Ames *et al.*, 1987). Standard practice in regulatory risk assessment for chemicals that induce tumors in high-dose rodent bioassays, has historically been to extrapolate risk to low dose in humans by multiplying rodent potency by human exposure, i.e. by assuming linearity in the dose response. Without data on mechanism of carcinogenesis, however, the true human risk of cancer at low dose is highly uncertain and could be zero (Ames and Gold, 1990; Clayson and Iverson, 1996; Gold *et al.*, 1992; Goodman, 1994). Adequate risk assessment from animal cancer tests requires more information for a chemical, about pharmacokinetics, mechanism of action, apoptosis, cell division, induction of defense and repair systems, and species differences. Several mechanisms have now been identified which indicate that exposures would not be expected to be a cancer risk to humans at the levels of usual exposure even though they induce tumors in high-dose tests (e.g. saccharin, BHA, chloroform, *d*-limonene). Under the new U.S. Environmental Protection Agency (EPA) Guidelines for Cancer Risk Assessment, these mechanisms are to be considered in evaluating the dose-response, method of risk assessment, and relevance to humans; the default linear extrapolation has been replaced by this more science-based approach (U.S. Environmental Protection Agency, 1999). Examples of such biologically-based mechanisms include cell proliferation following cytotoxic effects at high doses of saccharin, only in the male rat urothelium; the cytotoxicity results from formation of a precipitate in rat urine, which is a species-specific response. Several studies show an association between cell division in the rodent liver and cancer, e.g. chloroform, oxazepam, 2,4-diaminotoluene (Ames and Gold, 1990; Ames *et al.*, 1993a; Butterworth and Bogdanffy, 1999; Cohen, 1998; Cunningham *et al.*, 1994a; Cunningham *et al.*, 1991; Cunningham *et al.*, 1994b; Heddle, 1998). Some chemicals, e.g. *d*-limonene, induce kidney tumors in male rats by a mechanism which is not relevant to humans: accumulation of a male rat-specific protein ( $\alpha_{2u}$ -globulin) resulting in toxicity to the kidney, sustained cell proliferation, and kidney tumors. Humans do not synthesize  $\alpha_{2u}$ -globulin or any protein that can function like it (Swenberg and Lehman-McKeeman, 1999), and therefore the carcinogenic effect in male rats is not predictive of a cancer hazard to humans. Some chemicals induce thyroid follicular-cell tumors at high doses by a metabolic inactivation of the thyroid hormones  $T_3$  and  $T_4$  which results in increased levels of thyroid-stimulating hormone levels, sustained proliferation of cells in the thyroid, and tumor formation (McClain, 1990). Humans are less sensitive to this secondary, threshold mechanism than rats (McClain, 1994; U.S. Environmental Protection Agency, 1998a).

The U.S. EPA evaluation of chloroform provides an example of the new emphasis on incorporating more biological information into evaluations of cancer test results and risk assessment. The EPA concluded that chloroform-induced tumors were secondary to toxic effects that occur at high dose. Therefore EPA relied on a nonlinear dose-response approach with margin of exposure to estimate cancer risk for humans. They concluded that “chloroform is likely to be carcinogenic to humans by all routes of exposure under high-exposure conditions that lead to cytotoxicity and regenerative hyperplasia in susceptible tissues. Chloroform is not likely to be carcinogenic to humans by any route of exposure under exposure conditions that do not cause cytotoxicity and cell regeneration.” (U.S. Environmental Protection Agency, 2002)

### Is selection bias causing the high positivity rate?

Since the results of high-dose rodent tests are routinely used to identify a chemical as a possible cancer hazard to humans, it is important to try to understand how representative the 50% positivity rate might be of all untested chemicals. If half of all chemicals (both natural and synthetic) to which humans are exposed would be positive if tested, then the utility of a rodent bioassay to identify a chemical as a “potential human carcinogen” is questionable. To determine the true proportion of rodent carcinogens among chemicals would require a comparison of a random group of synthetic chemicals to a random group of natural chemicals. Such an analysis has not been done.

A counter argument to the idea that the 50% positivity rate is due to the effects of administering high doses, is that so many chemicals are positive because they were selected for testing on the grounds that they were expected to be carcinogenic. We have discussed that this is a likely bias since cancer testing is both expensive and time-consuming, making it prudent to test suspicious compounds (Gold *et al.*, 1998); however, chemicals are selected for cancer-testing for many reasons other than suspicion, including the extent of human exposure, level of production and occupational exposure, and scientific questions about carcinogenesis. Moreover, if the main basis for selection was that chemicals were suspected carcinogens, then one should select mutagens (80% are carcinogens compared to 49% of nonmutagens); yet, 55% of the chemicals tested are nonmutagens (Gold *et al.*, 1998). The idea that chemicals are selected for testing because they are likely to be carcinogenic, rests on an assumption that researchers have adequate knowledge about how to predict carcinogenicity, and that there is consensus about the criteria, i.e., the idea that bias in the positivity rate is due to selection requires that there is shared, adequate knowledge of what is likely to be carcinogenic. However, while some chemical classes are more often carcinogenic in rodent bioassays than others — for example, nitroso compounds, aromatic amines, nitroaromatics, and chlorinated compounds — several results suggest that predictive knowledge is highly imperfect, even now after decades of testing results have become available on which to base prediction. For example, a prospective prediction exercise was conducted by several experts in 1990 in advance of the 2-year bioassays by the United States National Toxicology Program (NTP). There was wide disagreement among the experts as to which chemicals would be carcinogenic when tested; accuracy varied, thus indicating that predictive knowledge is uncertain (Omenn *et al.*, 1995). One predictive analysis for a randomly selected group of chemicals has been conducted using a computerized method based on chemical structure; among 140 randomly selected chemicals, 65 (46%) were predicted to be carcinogenic if tested in standard bioassays (Rosenkranz and Klopman, 1990). Another argument against the selection bias hypothesis is the high positivity rate for drugs (Table 4), because drug development tends to select chemicals that are not mutagens or expected carcinogens.

A study by Innes *et al.* in 1969 (Innes *et al.*, 1969) has frequently been cited (Ames *et al.*, 1987, and Letters) as evidence that the positivity rate is low, because only 9% of 119 chemicals tested (primarily pesticides) were positive. However, the Innes tests were only in mice, had only 18 animals per group, and were terminated at 18 months. This protocol lacked the power of modern experiments, in which both rats and mice are tested, with 50 animals per group for 24 months. When 34 chemicals for which Innes obtained negative results were retested in other strains of mice or in rats, using more adequate protocols including higher doses and longer experiment length, 17 of the 34 formerly negative chemicals tested positive (Table 4) (Cohen, 1995; Cohen and Lawson, 1995; Gold *et al.*, 1999; Gold *et al.*, 1997a).

Thus it seems likely that a high proportion of all chemicals, whether synthetic or natural, might be “carcinogens” if run through the standard rodent bioassay at the MTD. For non-mutagens, carcinogenicity would be primarily due to the effects of high doses; for mutagens, it would result from a synergistic effect between cell division at high doses and DNA damage (Ames and Gold, 1990; Ames *et al.*, 1993a; Butterworth *et al.*, 1995). Without additional data on the mechanism of carcinogenesis for each chemical, the interpretation of a positive result in a rodent bioassay is highly uncertain. The carcinogenic effects may be limited to the high dose tested.

### **Problems in extrapolating carcinogenicity between species**

The use of bioassay results in risk assessment requires a qualitative species extrapolation from rats or mice to humans. The accuracy of this extrapolation is generally unverifiable, since data on humans are limited. Ultimately one wants to know whether the large number (many hundreds) of chemicals that have been shown to be carcinogenic in experimental animals would also be carcinogenic in humans. This question cannot be answered by reversing the question, i.e. by asking whether the small number of chemicals that are carcinogenic to humans are also carcinogenic in rodent bioassays. The reason for this is that even if most human carcinogens were carcinogenic to experimental animals, the converse does not necessarily follow, as can be demonstrated by a simple probabilistic argument (Freedman and Zeisel, 1988).

Evidence about interspecies extrapolation can, however, be obtained by investigating whether chemicals that are carcinogenic in rats are also carcinogenic in mice, and visa versa. If mice and rats are similar with respect to carcinogenesis, this provides some evidence in favor of interspecies extrapolations; conversely, if mice and rats are different, this casts doubt on the validity of extrapolations from mice to humans.

One measure of interspecies agreement is concordance, the percentage of chemicals that are classified the same way as to carcinogenicity in mice and rats (i.e. results are concordant if either a chemical is a carcinogen in both species or in neither, and results are discordant if a chemical is a carcinogen in one species but not in the other). Observed concordance in bioassays is about 75% (Gold *et al.*, 1997a; Gold *et al.*, 1998) which may seem low since the experimental conditions are identical and the species are similar. The observed concordance is just an estimate based on limited data. We have shown by simulations for 300 NCI/NTP bioassays of chemicals tested in both rats and mice, (which have an observed concordance of 75%), that an observed concordance of 75% can arise if the true concordance is anything between 20% and 100% (Freedman *et al.*, 1996; Lin *et al.*, 1995). In particular, observed concordance can seriously overestimate true concordance. Thus, it seems unlikely that true concordance between rats and mice can be estimated with any reasonable degree of confidence from bioassay data.

### **Problems in using animal cancer test results for regulatory risk assessment**

We have discussed validity problems associated with the use of the limited data from animal cancer tests for human risk assessment (Bernstein *et al.*, 1985; Gold *et al.*, 1998). Standard practice in regulatory risk assessment for a given rodent carcinogen has been to extrapolate from the high doses of rodent bioassays to the low doses of most human exposures by multiplying carcinogenic potency in rodents by human exposure. Strikingly, however, due to the relatively narrow range of doses in 2-year rodent bioassays and the limited range of statistically significant tumor incidence rates, the various measures of potency obtained from 2-year bioassays, such as the EPA  $q_1^*$  value, the  $TD_{50}$ , and the lower confidence limit on the  $TD_{10}$  ( $LTD_{10}$ ) are constrained



to a relatively narrow range of values about the MTD, in the absence of 100% tumor incidence at the target site, which rarely occurs (Bernstein *et al.*, 1985; Freedman *et al.*, 1993; Gaylor and Gold, 1995; Gaylor and Gold, 1998; Gold *et al.*, 1997a). For example, the dose usually estimated by regulatory agencies to give one cancer in a million, can be approximated simply by using the MTD as a surrogate for carcinogenic potency. Gaylor and Gold (1995) have shown that the “virtually safe dose” (VSD) can be approximated by the MTD/740,000 for rodent carcinogens tested in the bioassay program of the NCI/NTP. The MTD/740,000 was within a factor of 10 of the VSD for 96% of carcinogens. This is similar to the finding that in near-replicate experiments of the same chemical, potency estimates vary by a factor of 4 around a median value (Gaylor *et al.*, 1993; Gold *et al.*, 1989b; Gold *et al.*, 1987b).

Using the benchmark dose approach proposed in the EPA carcinogen guidelines, risk estimation is similarly constrained by bioassay design. A simple, quick, and relatively precise determination of the LTD<sub>10</sub> can be obtained by the maximum tolerated dose (MTD) divided by seven (Gaylor and Gold, 1998). Both linear extrapolation and the use of safety or uncertainty factors proportionately reduce a tumor dose in a similar manner. The difference in the regulatory “safe dose,” if any, for the two approaches depends on the magnitude of uncertainty factors selected. Using the benchmark dose approach of the proposed carcinogen risk assessment guidelines, the dose estimated from the LTD<sub>10</sub> divided, e.g., by a 1000-fold uncertainty factor is similar to the dose of an estimated risk of less than 10<sup>-4</sup> using a linear model. This dose is 100 times higher than the VSD corresponding to an estimated risk of less than 10<sup>-6</sup>. Thus, whether the procedure involves a benchmark dose or a linearized model, cancer risk estimation is constrained by the bioassay design.

### **Misconception 7—Synthetic chemicals pose greater carcinogenic hazards than natural chemicals**

An analysis of synthetic chemicals in the perspective of the vast array of natural chemicals shows that synthetic rodent carcinogens are a tiny fraction of the total. In several papers (Ames *et al.*, 1995; Ames *et al.*, 1987; Ames *et al.*, 1990a; Gold *et al.*, 1999; Gold *et al.*, 1992), we have emphasized the importance of setting research and regulatory priorities by gaining a broad perspective about the vast number of chemicals to which humans are exposed. A comparison of potential hazards using a simple index can be helpful in efforts to communicate what might be important factors in cancer prevention and when selecting chemicals for *chronic bioassay*, mechanistic, or epidemiologic studies (Ames *et al.*, 1987; Ames *et al.*, 1990b; Gold *et al.*, 1992; Gold and Zeiger, 1997). There is a need to identify what might be the important cancer hazards among the ubiquitous exposures to rodent carcinogens in everyday life.

### **Human Exposure/Rodent Potency (HERP)—ranking possible cancer hazards from rodent carcinogens**

One reasonable strategy for setting priorities is to use a rough index to compare and rank possible carcinogenic hazards from a wide variety of chemical exposures at levels that humans typically receive, and then to focus on those that rank highest (Gold *et al.*, 1999; Gold *et al.*, 1997a; Gold *et al.*, 1992). Ranking is thus a critical first step. Although one cannot say whether the ranked chemical exposures are likely to be of major or minor importance in human cancer, it is not prudent to focus attention on the possible hazards at the bottom of a ranking if, by using the same methodology to identify hazard, there are numerous common human exposures with much

greater possible hazards. Research on mechanism of carcinogenesis for a given chemical is needed to interpret the possible human risk. Our analyses are based on the Human Exposure/Rodent Potency index (HERP) which indicates what percentage of the rodent carcinogenic potency ( $TD_{50}$  in mg/kg/day) a person receives from a given average daily dose when exposed over a lifetime (mg/kg/day) (Gold and Zeiger, 1997) [Table 5]. The method for calculating the HERP index, including an example, is described in the Appendix.  $TD_{50}$  values in our CPDB span a 10 million-fold range across chemicals (Gold *et al.*, 1997c). Human exposures to rodent carcinogens range enormously as well, from historically high workplace exposures in some occupations or pharmaceutical dosages to very low exposures from residues of synthetic chemicals in food or water. Consideration of both these values for a chemical is necessary for ranking possible hazard.

Overall, our HERP ranking has shown that synthetic pesticide residues rank low in possible carcinogenic hazard compared to many common exposures. HERP values for some historically high exposures in the workplace and some pharmaceuticals rank high, and there is an enormous background of naturally-occurring rodent carcinogens in average consumption of common foods. This background of natural chemical results casts doubt on the relative importance of low-dose exposures to residues of synthetic chemicals such as pesticides (Ames *et al.*, 1987; Gold *et al.*, 1994a; Gold *et al.*, 1992). A committee of the National Research Council recently reached similar conclusions about natural vs. synthetic chemicals in the diet, and called for further research on natural chemicals (National Research Council, 1996). The rank order of possible hazards by HERP is similar to the order that would be based on a linear model.

The ranking of possible hazards (HERP values in %) in Table 5 is for *average* United States exposures to all rodent carcinogens in the CPDB for which concentration data and average United States exposure or consumption data were both available, and for which human exposure could be chronic for a lifetime. For pharmaceuticals the doses are recommended doses, and for workplace they are past industry or occupation averages. The 94 exposures in the ranking [Table 5] are ordered by possible carcinogenic hazard (HERP), and natural chemicals in the diet are reported in boldface. Several HERP values make convenient reference points for interpreting Table 5. The median HERP value is 0.002%, and the background HERP for the average chloroform level in a liter of United States tap water is 0.0008%. Chloroform is formed as a by-product of water chlorination, and the HERP value reflects exposure to chloroform from both drinking water and breathing indoor air e.g. when showering (chloroform is volatile.). A HERP of 0.00001% is approximately equal to a regulatory risk level of 1-in-a-million based on a linear model, i.e. the Virtually Safe Dose (VSD) (Gold *et al.*, 1992). The rank order in Table 5 would be the same for a Margin of Exposure from the  $TD_{50}$  because the MOE is inversely related to HERP. Table 5 indicates that if the same methodology is used for both naturally-occurring and synthetic chemicals, most ordinary foods would not pass the default regulatory criteria that have been used for synthetic chemicals. For many natural chemicals the HERP values are in the top half of the table, even though natural chemicals are markedly underrepresented because so few have been tested in rodent bioassays. The ranking of HERP values maximizes possible hazards from synthetic chemicals because it includes historically high exposure values that are now much lower: for example, exposure to DDT and saccharin as well as to occupational chemicals.

For readers who are interested in the results for particular categories of exposure or particular chemicals, we discuss below several categories of exposure and selected chemicals. We indicate for some chemicals the mechanistic data suggesting that the rodent results may not be

relevant to humans or that possible hazards would be lower if nonlinearity or a threshold in the dose-response were taken into account in risk assessment.

**Table 5: Ranking possible carcinogenic hazards from average US exposures to rodent carcinogens**

Chemicals that occur naturally in food are in green.

**Possible hazard HERP (%)** (Column 1) is calculated using the information in columns 2, 3, and 4. **Average daily US human exposure** (Column 2) indicates a daily dose for a lifetime for drugs, the air in the workplace or home, food, water, residues etc. **Human dose of rodent carcinogen** (Column 3) is divided by 70 kg to give a mg/kg/day of human exposure. The **Human Exposure/Rodent Potency index (HERP)** in Column 1 expresses this human dose as a percentage of the **TD<sub>50</sub>** in the rodent (mg/kg/day), which is reported in Column 4. TD<sub>50</sub> values used in the HERP calculation are averages calculated by taking the harmonic mean of the TD<sub>50</sub>s of the positive tests in that species from the Carcinogenic Potency Database (CPDB). Average TD<sub>50</sub> values have been calculated separately for rats and mice, and the more potent value is used for calculating possible hazard. (See **Appendix**, below for more details.) “.” = no data in CPDB; a number in parentheses indicates a TD<sub>50</sub> value not used in the HERP calculation because TD<sub>50</sub> is less potent than in the other species. (-) = negative in cancer tests; (+) = positive cancer test(s) not suitable for calculating a TD<sub>50</sub>.

Possible hazard: HERP (%)	Average daily U.S. exposure	Human dose of rodent carcinogen	Potency TD <sub>50</sub> (mg/kg/day) <sup>a</sup>		Exposure references
			Rats	Mice	
140	EDB: production workers (high exposure) (before 1977)	Ethylene dibromide, 150 mg	1.52	(7.45)	(Ott <i>et al.</i> , 1980; Ramsey <i>et al.</i> , 1978)
17	Clofibrate	Clofibrate, 2 g	169	.	(Havel and Kane, 1982)
12	Phenobarbital, 1 sleeping pill	Phenobarbital, 60 mg	(+)	7.38	(American Medical Association Division of Drugs, 1983)
6.9	Gemfibrozil	Gemfibrozil, 1.2 g	247	(-)	(Arky, 1998)
6.8	Styrene-butadiene rubber industry workers (1978-86)	1,3-Butadiene, 66.0 mg	(261)	13.9	(Matanoski <i>et al.</i> , 1993)
6.2	<b>Comfrey-pepsin tablets, 9 daily (no longer recommended)</b>	<b>Comfrey root, 2.7 g</b>	626	.	(Culvenor <i>et al.</i> , 1980; Hirono <i>et al.</i> , 1978)
6.1	Tetrachloroethylene: dry cleaners with dry-to-dry units (1980-90)	Tetrachloroethylene, 433 mg	101	(126)	(Andrasik and Cloutet, 1990)
4.0	Formaldehyde: production workers (1979)	Formaldehyde, 6.1 mg	2.19	(43.9)	(Siegal <i>et al.</i> , 1983)
3.6	<b>Alcoholic beverages, all types</b>	<b>Ethyl alcohol, 22.8 ml</b>	9110	(-)	(Nephew <i>et al.</i> , 2000)
2.4	Acrylonitrile: production workers (1960-1986)	Acrylonitrile, 28.4 mg	16.9	.	(Blair <i>et al.</i> , 1998)
2.2	Trichloroethylene: vapor degreasing (before 1977)	Trichloroethylene, 1.02 g	668	(1580)	(Page and Arthur, 1978)
1.8	<b>Beer, 229 g</b>	<b>Ethyl alcohol, 11.7 ml</b>	9110	(-)	(Beer Institute, 1999)
1.4	Mobile home air (14 hours/day)	Formaldehyde, 2.2 mg	2.19	(43.9)	(Connor <i>et al.</i> , 1985)
1.3	<b>Comfrey-pepsin tablets, 9 daily (no longer recommended)</b>	<b>Symphytine, 1.8 mg</b>	1.91	.	(Culvenor <i>et al.</i> , 1980; Hirono <i>et al.</i> , 1978)

Possible hazard: HERP (%)	Average daily U.S. exposure	Human dose of rodent carcinogen	Potency TD <sub>50</sub> (mg/kg/day) <sup>a</sup>		Exposure references
			Rats	Mice	
0.9	Methylene chloride: workers, industry average (1940s-80s)	Methylene chloride, 471 mg	724	(1100)	(CONSAD Research Corporation, 1990)
0.6	<b>Wine, 20.8 g</b>	<b>Ethyl alcohol, 3.67 ml</b>	9110	(-)	(Wine Institute, 2001)
0.5	Dehydroepiandrosterone (DHEA)	DHEA supplement, 25 mg	68.1	.	
0.4	Conventional home air (14 hours/day)	Formaldehyde, 598 µg	2.19	(43.9)	(McCann <i>et al.</i> , 1987)
0.2	Fluvastatin	Fluvastatin, 20 mg	125	.	(Arky, 1998)
0.1	<b>d-Limonene in food</b>	<b>d-Limonene, 15.5 mg</b>	204	(-)	(Stofberg and Grundschober, 1987)
0.1	<b>Coffee, 11.6 g</b>	<b>Caffeic acid, 20.8 mg</b>	297	(4900)	(Clarke and Macrae, 1988; Coffee Research Institute, 2001)
0.06	Lovastatin	Lovastatin, 20 mg	(-)	515	(Arky, 1998)
0.04	<b>Lettuce, 14.9 g</b>	<b>Caffeic acid, 7.90 mg</b>	297	(4900)	(Herrmann, 1978; Technical Assessment Systems, 1989)
0.03	<b>Safrole in spices</b>	<b>Safrole, 1.2 mg</b>	(441)	51.3	(Hall <i>et al.</i> , 1989)
0.03	<b>Orange juice, 138 g</b>	<b>d-Limonene, 4.28 mg</b>	204	(-)	(Schreier <i>et al.</i> , 1979; Technical Assessment Systems, 1989)
0.03	<b>Comfrey herb tea, 1 cup (1.5 g root) (no longer recommended)</b>	<b>Symphytine, 38 µg</b>	1.91	.	(Culvenor <i>et al.</i> , 1980)
0.03	<b>Tomato, 88.7 g</b>	<b>Caffeic acid, 5.46 mg</b>	297	(4900)	(Schmidtlein and Herrmann, 1975a; Technical Assessment Systems, 1989)
0.03	<b>Furfural in food</b>	<b>Furfural, 3.64 mg</b>	(683)	197	(Adams <i>et al.</i> , 1997)
0.02	<b>Coffee, 11.6 g</b>	<b>Catechol, 1.16 mg</b>	84.7	(244)	(Coffee Research Institute, 2001; Rahn and König, 1978; Tressl <i>et al.</i> , 1978)
0.02	<b>Mushroom (<i>Agaricus bisporus</i> 2.55 g)</b>	<b>Mixture of hydrazines, etc. (whole mushroom)</b>	(-)	20,300	(Matsumoto <i>et al.</i> , 1991; Stofberg and Grundschober, 1987; Toth and Erickson, 1986)
0.02	<b>Apple, 32.0 g</b>	<b>Caffeic acid, 3.40 mg</b>	297	(4900)	(Mosel and Herrmann, 1974; U.S. Environmental Protection Agency. Office of Pesticide Programs, 1989)
0.01	BHA: daily U.S. avg (1975)	BHA, 4.6 mg	606	(5530)	(U.S. Food and Drug Administration, 1991a)
0.01	<b>Beer (before 1979), 229 g</b>	<b>Dimethylnitrosamine, 646 ng</b>	0.0959	(0.189)	(Beer Institute, 1999; Fazio <i>et al.</i> , 1980; Preussmann and Eisenbrand, 1984)
0.008	<b>Aflatoxin: daily U.S. avg (1984-89)</b>	<b>Aflatoxin, 18 ng</b>	0.0032	(+)	(U.S. Food and Drug Administration, 1992)
0.007	<b>Celery, 14 g</b>	<b>Caffeic acid, 1.51 mg</b>	297	(4900)	(Smiciklas-Wright <i>et al.</i> , 2002; Stöhr and Herrmann, 1975)
0.007	d-Limonene	Food additive, 1.01 mg	204	(-)	(Lucas <i>et al.</i> , 1999)
0.007	<b>Cinnamon, 21.9 mg</b>	<b>Coumarin, 65.0 µg</b>	13.9	(103)	(Poole and Poole, 1994)
0.006	<b>Coffee, 11.6 g</b>	<b>Furfural, 783 µg</b>	(683)	197	(Coffee Research Institute, 2001; Stofberg and Grundschober, 1987)
0.005	<b>Coffee, 11.6 g</b>	<b>Hydroquinone, 290 µg</b>	82.8	(225)	(Coffee Research Institute, 2001; Heinrich and Baltes, 1987; Tressl <i>et al.</i> , 1978)
0.005	Saccharin: daily U.S. avg (1977)	Saccharin, 7 mg	2140	(-)	(National Research Council, 1979)
0.005	<b>Carrot, 12.1 g</b>	<b>Aniline, 624 µg</b>	194 <sup>b</sup>	(-)	(Neurath <i>et al.</i> , 1977; Technical Assessment Systems, 1989)

Possible hazard: HERP (%)	Average daily U.S. exposure	Human dose of rodent carcinogen	Potency TD <sub>50</sub> (mg/kg/day) <sup>a</sup>		Exposure references
			Rats	Mice	
0.004	<b>Bread, 79 g</b>	<b>Furfural, 584 µg</b>	(683)	197	(Smiciklas-Wright <i>et al.</i> , 2002; Stofberg and Grundschober, 1987)
0.004	<b>Potato, 54.9 g</b>	<b>Caffeic acid, 867 µg</b>	297	(4900)	(Schmidlein and Herrmann, 1975b; Technical Assessment Systems, 1989)
0.004	<b>Methyl eugenol in food</b>	<b>Methyl eugenol, 46.2 µg</b>	(19.7)	18.6	(Smith <i>et al.</i> , 2002)
0.003	Conventional home air (14 hour/day)	Benzene, 155 µg	(169)	77.5	(McCann <i>et al.</i> , 1987)
0.002	<b>Coffee, 11.6 g</b>	<b>4-Methylcatechol, 378 µg</b>	248	.	(Coffee Research Institute, 2001; Heinrich and Baltes, 1987; International Agency for Research on Cancer, 1991)
0.002	<b>Nutmeg, 17.6 mg</b>	<b>d-Limonene, 299 µg</b>	204	(-)	(Bejnarowicz and Kirch, 1963; U.S. Department of Agriculture, 2000)
0.002	<b>Carrot, 12.1 g</b>	<b>Caffeic acid, 374 µg</b>	297	(4900)	(Stöhr and Herrmann, 1975; Technical Assessment Systems, 1989)
0.002	Ethylene thiourea: daily U.S. avg (1990)	Ethylene thiourea, 9.51 µg	7.9	(23.5)	(U.S. Environmental Protection Agency, 1991a)
0.002	BHA: daily U.S. avg (1987)	BHA, 700 µg	606	(5530)	(U.S. Food and Drug Administration, 1991a)
0.002	DDT: daily U.S. avg (before 1972 ban) <sup>c</sup>	DDT, 13.8 µg	(84.7)	12.8	(Duggan and Corneliussen, 1972)
0.001	<b>Estragole in spices</b>	<b>Estragole, 54.0 µg</b>	.	51.8	(Smith <i>et al.</i> , 2002)
0.001	<b>Pear, 3.7 g</b>	<b>Caffeic acid, 270 µg</b>	297	(4900)	(Mosel and Herrmann, 1974; U.S. Environmental Protection Agency, 1997)
0.001	Toxaphene: daily U.S. avg (before 1982 ban) <sup>c</sup>	Toxaphene, 6.43 µg	(-)	7.51	(Podrebarac, 1984)
0.001	<b>Mushroom (<i>Agaricus bisporus</i> 5.34 g)</b>	<b>Glutamyl-p-hydrazino-benzoate, 224 µg</b>	.	277	(Chauhan <i>et al.</i> , 1985; U.S. Food and Drug Administration, 2002)
0.001	<b>Plum, 1.7 g</b>	<b>Caffeic acid, 235 µg</b>	297	(4900)	(Mosel and Herrmann, 1974; U.S. Environmental Protection Agency, 1997)
0.001	[UDMH: daily U.S. avg (1988)]	[UDMH, 2.82 µg (from Alar)]	(-)	3.96	(U.S. Environmental Protection Agency. Office of Pesticide Programs, 1989)
0.001	<b>Bacon, 19 g</b>	<b>Diethylnitrosamine, 19 ng</b>	0.0266	(+)	(Sen <i>et al.</i> , 1979; Smiciklas-Wright <i>et al.</i> , 2002)
0.0008	<b>Bacon, 19 g</b>	<b>Dimethylnitrosamine, 57.0 ng</b>	0.0959	(0.189)	(Smiciklas-Wright <i>et al.</i> , 2002; Tricker and Preussmann, 1991)
0.0008	Tap water, 1 liter (1987-92)	Chloroform, 51 µg	(262)	90.3	(American Water Works Association. Government Affairs Office, 1993; McKone, 1987; McKone, 1993)
0.0008	DDE: daily U.S. avg (before 1972 ban) <sup>c</sup>	DDE, 6.91 µg	(-)	12.5	(Duggan and Corneliussen, 1972)
0.0007	TCDD: daily U.S. avg (1994)	TCDD, 12.0 pg	0.0000235	(0.000156)	(U.S. Environmental Protection Agency, 2000)
0.0007	<b>Bacon, 19 g</b>	<b>N-Nitrosopyrrolidine, 324 ng</b>	(0.799)	0.679	(Stofberg and Grundschober, 1987; Tricker and Preussmann, 1991)
0.0006	Methyl eugenol	Food additive, 7.7 µg	(19.7)	18.6	(Smith <i>et al.</i> , 2002)
0.0004	EDB: Daily U.S. avg (before 1984 ban) <sup>c</sup>	EDB, 420 ng	1.52	(7.45)	(U.S. Environmental Protection Agency. Office of Pesticide Programs, February 8, 1984)
0.0004	Tap water, 1 liter (1987-92)	Bromodichloromethane, 13 µg	(72.5)	47.7	(American Water Works Association. Government Affairs Office, 1993)

Possible hazard: HERP (%)	Average daily U.S. exposure	Human dose of rodent carcinogen	Potency TD <sub>50</sub> (mg/kg/day) <sup>a</sup>		Exposure references
			Rats	Mice	
0.0004	<b>Celery, 14 g</b>	<b>8-Methoxypsoralen, 8.56 µg</b>	32.4	(-)	(Beier <i>et al.</i> , 1983; Smiciklas-Wright <i>et al.</i> , 2002)
0.0003	<b>Mango, 1.0 g</b>	<b>d-Limonene, 40.0 µg</b>	204	(-)	(Engel and Tressl, 1983; U.S. Environmental Protection Agency, 1997)
0.0003	Furfural	Food additive, 36.4 µg	(683)	197	(Lucas <i>et al.</i> , 1999)
0.0003	Carbaryl: daily U.S. avg (1990)	Carbaryl, 2.6 µg	14.1	(-)	(U.S. Food and Drug Administration, 1991b)
0.0003	<b>Mustard, 18.9 mg</b>	<b>Allyl isothiocyanate, 17.4 µg</b>	96	(-)	(Krul <i>et al.</i> , 2002; Lucas <i>et al.</i> , 1999; Tsao <i>et al.</i> , 2002)
0.0002	<b>Beer (1994-95), 229 g</b>	<b>Dimethylnitrosamine, 16 ng</b>	0.0959	(0.189)	(Beer Institute, 1999; Glória <i>et al.</i> , 1997)
0.0002	<b>Mushroom (<i>Agaricus bisporus</i>, 5.34 g)</b>	<b>p-Hydrazinobenzoate, 58.6 µg</b>	.	454 <sup>b</sup>	(Chauhan <i>et al.</i> , 1985; U.S. Food and Drug Administration, 2002)
0.0002	Estragole	Food additive, 5.79 µg	.	51.8	(Lucas <i>et al.</i> , 1999)
0.0002	Allyl isothiocyanate	Food additive, 10.5 µg	96	(-)	(Lucas <i>et al.</i> , 1999)
0.0002	<b>Hamburger, pan fried, 85 g</b>	<b>PhIP, 176 ng</b>	1.64 <sup>b</sup>	(28.6 <sup>b</sup> )	(Knize <i>et al.</i> , 1994; Technical Assessment Systems, 1989)
0.0001	Toxaphene: daily U.S. avg (1990) <sup>c</sup>	Toxaphene, 595 ng	(-)	7.51	(U.S. Food and Drug Administration, 1991b)
0.00008	PCBs: daily U.S. avg (1984-86)	PCBs, 98 ng	1.74	(9.58)	(Gunderson, 1995)
0.00008	<b>Toast, 79 g</b>	<b>Urethane, 948 ng</b>	(41.3)	16.9	(Canas <i>et al.</i> , 1989; Smiciklas-Wright <i>et al.</i> , 2002)
0.00008	DDE/DDT: daily U.S. avg (1990) <sup>c</sup>	DDE, 659 ng	(-)	12.5	(U.S. Food and Drug Administration, 1991b)
0.00007	<b>Beer, 229 g</b>	<b>Furfural, 9.50 µg</b>	(683)	197	(Beer Institute, 1999; Lau and Lindsay, 1972; Tressl, 1976; Wheeler <i>et al.</i> , 1971)
0.00006	<b>Parsnip, 48.8 mg</b>	<b>8-Methoxypsoralen, 1.42 µg</b>	32.4	(-)	(Ivie <i>et al.</i> , 1981; U.S. Environmental Protection Agency, 1997)
0.00004	<b>Parsley, fresh, 257 mg</b>	<b>8-Methoxypsoralen, 928 ng</b>	32.4	(-)	(Chaudhary <i>et al.</i> , 1986; U.S. Environmental Protection Agency, 1997)
0.00003	<b>Hamburger, pan fried, 85 g</b>	<b>MeIQx, 38.1 ng</b>	1.66	(24.3)	(Knize <i>et al.</i> , 1994; Technical Assessment Systems, 1989)
0.00002	Dicofol: daily U.S. avg (1990)	Dicofol, 544 ng	(-)	32.9	(U.S. Food and Drug Administration, 1991b)
0.00001	<b>Hamburger, pan fried, 85 g</b>	<b>IQ, 6.38 ng</b>	0.921 <sup>b</sup>	(19.6)	(Knize <i>et al.</i> , 1994; Technical Assessment Systems, 1989)
0.000009	<b>Beer, 229 g</b>	<b>Urethane, 102 ng</b>	(41.3)	16.9	(Beer Institute, 1999; Canas <i>et al.</i> , 1989)
0.000005	Hexachlorobenzene: daily U.S. avg (1990)	Hexachlorobenzene, 14 ng	3.86	(65.1)	(U.S. Food and Drug Administration, 1991b)
0.000001	Lindane: daily U.S. avg (1990)	Lindane, 32 ng	(-)	30.7	(U.S. Food and Drug Administration, 1991b)
0.0000004	PCNB: daily U.S. avg (1990)	PCNB (Quintozene), 19.2 ng	(-)	71.1	(U.S. Food and Drug Administration, 1991b)
0.0000001	Chlorobenzilate: daily U.S. avg (1989) <sup>c</sup>	Chlorobenzilate, 6.4 ng	(-)	93.9	(U.S. Food and Drug Administration, 1991b)
0.00000008	Captan: daily U.S. avg (1990)	Captan, 115 ng	2080	(2110)	(U.S. Food and Drug Administration, 1991b)
0.00000001	Folpet: daily U.S. avg (1990)	Folpet, 12.8 ng	(-)	1550	(U.S. Food and Drug Administration, 1991b)
<0.00000001	Chlorothalonil: daily U.S. avg (1990)	Chlorothalonil, <6.4 ng	828 <sup>d</sup>	(-)	(U.S. Environmental Protection Agency, 1987; U.S. Food and Drug Administration, 1991b)

<sup>a</sup> "." = no data in CPDB; a number in parentheses indicates a TD<sub>50</sub> value not used in the HERP calculation because TD<sub>50</sub> is less potent than in the other species.

(-) = negative in cancer tests; (+) = positive cancer test(s) not suitable for calculating a TD<sub>50</sub>.

<sup>b</sup> TD<sub>50</sub> harmonic mean was estimated for the base chemical from the hydrochloride salt.

<sup>c</sup> No longer contained in any registered pesticide product (USEPA, 1998).

<sup>d</sup> Additional data from the EPA that is not in the CPDB were used to calculate this TD<sub>50</sub> harmonic mean.



## Occupational exposures

Occupational exposures to some chemicals have been high, and many of the single chemical agents or industrial processes evaluated as human carcinogens have been identified by historically high exposures in the workplace (International Agency for Research on Cancer, 1971-2002; Tomatis and Bartsch, 1990). HERP values rank at or near the top of Table 5 for highly exposed occupational groups, mostly from the past: ethylene dibromide, 1,3-butadiene, tetrachloroethylene, formaldehyde, acrylonitrile, trichloroethylene, and methylene chloride. The assessment of exposure in occupational settings is often difficult because workers are often exposed occupationally to more than one chemical at a time or over the course of a worklife. Epidemiological studies are often small and lack information on potentially confounding factors such as smoking and alcohol consumption. IARC has evaluated the evidence in humans as limited for butadiene, trichloroethylene, tetrachloroethylene, and formaldehyde; for ethylene dibromide, acrylonitrile, and methylene chloride the evidence is inadequate (International Agency for Research on Cancer, 1971-2002). Unlike the IARC, the National Toxicology Program (U.S. National Toxicology Program, 2000b) considered 1,3-butadiene to be a human carcinogen; the two agencies differed with respect to their evaluation of the strength of evidence for leukemia in butadiene-exposed workers and in whether an increased risk in the styrene-butadiene industry may have been due to exposures other than butadiene (International Agency for Research on Cancer, 1999a; U.S. National Toxicology Program, 2000b). The rodent carcinogens listed in the HERP table as occupational exposures also occur naturally, with the exception of ethylene dibromide; for example, butadiene occurs in forest fires, environmental tobacco smoke, and heated cooking oils (Shields *et al.*, 1995); acrylonitrile occurs in cigarette smoke; formaldehyde is ubiquitous in food, is generated metabolically in animals, and is present in human blood.

The possible hazard estimated for past actual exposure levels of the most heavily exposed EDB workers is the highest in Table 5 (HERP=140%). We testified in 1981 that our calculations showed that the workers were allowed to breathe in a dose higher than the dose that gave half of the test rats cancer, although the level of human exposure may have been somewhat overestimated (California Department of Health Services, 1985). An epidemiologic study of these workers, who inhaled EDB for over a decade, did not show any increase in cancer; however, because of the relatively small numbers of people the study lacked the statistical power to detect a small effect (California Department of Health Services, 1985; Ott *et al.*, 1980; Ramsey *et al.*, 1978). Ethylene dibromide is no longer produced in the U.S., and nearly all of its uses have been discontinued (the primary use was as an antiknock agent in leaded gasoline).

For trichloroethylene (TCE), the HERP is 2.2% for workers who cleaned equipment with TCE prior to 1977 (vapor degreasers). We recently conducted an analysis (Bogen and Gold, 1997) based on the assumption that carcinogenic effects are due to toxic effects from peak doses to the liver, the target organ for trichloroethylene carcinogenicity in mice. Our estimates indicate that for occupational respiratory exposures, the Permissible Exposure Limit (PEL) for trichloroethylene would produce concentrations of TCE metabolites that are higher than the no observed effect level (NOEL) for liver toxicity in mice. On this basis, the PEL is not expected to be protective. In contrast, the EPA maximum concentration limit (MCL) in drinking water of 5  $\mu\text{g/liter}$  based on a linearized multistage model, is more stringent than our safe dose estimate based on a 1000-fold safety factor, which is 210  $\mu\text{g/liter}$  (Bogen and Gold, 1997).

In other analyses, we used PELs of the United States Occupational Safety and Health Administration (OSHA), as surrogates for actual exposures and compared the permitted daily dose-rate for workers with the TD<sub>50</sub> in rodents (PERP index, Permissible Exposure/Rodent Potency) (Gold *et al.*, 1987a; Gold *et al.*, 1994a) For current permitted levels, PERP values for 14 chemicals are greater than 10%. Because workers can be exposed chronically to high doses of chemicals, it is important to have protective exposure limits (Gold *et al.*, 1994a). In recent years the permitted exposures for 1,3-butadiene and methylene chloride have been lowered substantially in the U.S., and the current PERP values are below 1%.

### **Pharmaceuticals and herbal supplements**

In Table 4 we reported that half the drugs in the Physician's Desk Reference (PDR) that have reported cancer test data, are carcinogens in rodent bioassays (Davies and Monro, 1995), as are 44% of drug submissions to United States Food and Drug Administration (FDA) (Contrera *et al.*, 1997). Most drugs, however, are used for only short periods, and therefore we have not calculated HERP values for them. Pharmaceuticals are evaluated by the FDA using mechanistic data as well as tumor incidence, and taking benefits into account.

The HERP ranking includes pharmaceuticals that can be used chronically; some are high in the HERP ranking, primarily because the dose ingested is high. Phenobarbital (HERP =12%) is a sedative and anticonvulsant that has been investigated in humans who took it for decades; there is no convincing evidence that it caused cancer (American Medical Association Division of Drugs, 1983; Freidman and Habel, 1999; McLean *et al.*, 1986). Mechanistic data suggest that the dose-response curve for tumors induced in rodents is nonlinear and perhaps, exhibits a threshold. Four cholesterol-lowering drugs have evidence of carcinogenicity in rodent tests; they are not mutagenic or genotoxic, and long-term epidemiological studies and clinical trials have not provided evidence of an association with fatal or non-fatal cancers in humans (Bjerre and LeLorier, 2001; Childs and Girardot, 1992; Havel and Kane, 1982; International Agency for Research on Cancer, 1996; Pfeffer *et al.*, 2002; Reddy and Lalwani, 1983; World Health Organization, 1984). Two of these drugs, clofibrate (HERP = 17%) which was used as a cholesterol lowering agent primarily before the 1970s, and gemfibrozil (HERP = 6.9%) which is currently used, increase liver tumors in rodents by the mechanism of peroxisome proliferation, which suggests that they would not be expected to be carcinogenic in humans (Cattley *et al.*, 1996; Havel and Kane, 1982; Reddy and Lalwani, 1983; World Health Organization, 1984). The two other cholesterol-lowering drugs in Table 6 are statins: fluvastatin (HERP=0.2%) and the widely-used drug, lovastatin (HERP=0.06%). Large clinical trials of statins have shown no carcinogenic effects in humans, although there were limitations in the studies: the follow-up period of 5 years is short for observing carcinogenic effects and the trials were not designed to measure cancer risk (Bjerre and LeLorier, 2001; Guallar and Goodman, 2001; Pfeffer *et al.*, 2002). A meta-analysis of 5 clinical trials examined only the combination of all cancers rather than specific types of cancer (Guallar and Goodman, 2001).

Herbal supplements have recently developed into a large market in the United States; they have not been a focus of carcinogenicity testing. The FDA regulatory requirements for safety and efficacy that are applied to pharmaceuticals do not pertain to herbal supplements under the 1994 Dietary Supplement and Health Education Act (DSHEA), and few have been tested for carcinogenicity. The relevant regulatory requirements in Canada are under review, and current regulations treat non-prescription ingredients of botanical origin separately from pharmaceuticals (Health Canada, 1995; Volpe, 1998). Those that are rodent carcinogens tend to rank

high in HERP because, like some pharmaceutical drugs, the recommended dose is high relative to the rodent carcinogenic dose. Moreover, under DSHEA the safety criteria that have been used for decades by FDA for food additives that are “Generally Recognized As Safe” (GRAS) are not applicable to dietary supplements (Burdock, 2000) even though supplements are used at higher doses. The NTP is currently testing several medicinal herbs or chemicals that are present in herbs.

### *Comfrey*

Comfrey is a medicinal herb whose roots and leaves have been shown to be carcinogenic in rats. For the formerly recommended dose of 9 daily comfrey-pepsin tablets HERP=6.2%. Symphytine, a pyrrolizidine-alkaloid which is a natural plant pesticide, is a rodent carcinogen present in comfrey-pepsin tablets and comfrey tea. The HERP value for symphytine is 1.3% in the pills and 0.03% in comfrey herb tea. Comfrey pills are no longer widely sold, but are available on the World Wide Web. Comfrey roots and leaves can be bought at health food stores and on the Web and can thus be used for tea, although comfrey is recommended for topical use only in the *PDR for Herbal Medicines* (Gruenwald *et al.*, 1998). Poisoning epidemics by pyrrolizidine alkaloids have occurred in the developing world. In the United States poisonings, including deaths, have been associated with use of herbal teas containing comfrey (Huxtable, 1995). Recently the U.S. FDA issued a warning about comfrey and asked manufacturers to withdraw their comfrey products after several people became ill from taking comfrey as a supplement or as tea. Comfrey is banned from distribution in Canada (Stickel and Seitz, 2000). Several other pyrrolizidine-containing medicinal plants are rodent carcinogens, including coltsfoot, *Senecio longilobus* and *S. nemorensis*, *Petasites japonicus*, and *Farfugium japonicum*. Over 200 pyrrolizidine alkaloids are present in more than 300 plant species. Up to 3% of flowering plant species contain pyrrolizidine alkaloids (Prakash *et al.*, 1999). Several pyrrolizidine alkaloids have been tested chronically in rodent bioassays and are carcinogenic (Gold *et al.*, 1997c).

### *Dehydroepiandrosterone (DHEA)*

Dehydroepiandrosterone (DHEA) and DHEA sulfate are the major secretion products of adrenal glands in humans and are precursors of androgenic and estrogenic hormones (Oelkers, 1999; van Vollenhoven, 2000). DHEA is manufactured as a dietary supplement, and sold widely for a variety of purposes including the delay of aging. DHEA is a controlled drug in Canada (Health Canada, 2000). In rats, DHEA induces liver tumors (Hayashi *et al.*, 1994; Rao *et al.*, 1992), and the HERP value for the recommended human dose of one daily capsule containing 25 mg DHEA is 0.5%. Peroxisome proliferation is the mechanism of liver carcinogenesis in rats for DHEA, suggesting that the carcinogenicity may not be relevant to humans (Hayashi *et al.*, 1994). DHEA inhibited the development of tumors of the rat testis (Rao, 1992) and the rat and mouse mammary gland (McCormick *et al.*, 1996; Schwartz *et al.*, 1981). A recent review of clinical, experimental, and epidemiological studies concluded that late promotion of breast cancer in postmenopausal women may be stimulated by prolonged intake of DHEA (Stoll, 1999); however the evidence for a positive association in postmenopausal women between serum DHEA levels and breast cancer risk is conflicting (Bernstein *et al.*, 1990; Stoll, 1999).

### *Aristolochic acid*

Herbal medicinal products containing aristolochic acid have been found to induce urinary tract cancer in humans, and the FDA has issued warnings about supplements and traditional medicines that contain aristolochic acid (Schwetz, 2001, <http://www.cfsan.fda.gov/%20~dms/ds-bot.html>).

*Aristolochia* species, which are the source of aristolochic acid, are listed in the Chinese pharmacopoeia (Reid, 1993). In a diet clinic in Belgium, aristolochic acid was unintentionally administered to patients in pills which purportedly contained a chemical from a different plant species. Many of the female patients who took aristolochic acid developed kidney disease (***Chinese-herb nephropathy***), and the cumulative dose of aristolochic acid was related to the progression of the disease. Thirty nine patients suffered terminal renal failure, and of these, 18 developed urothelial tract carcinoma (Nortier *et al.*, 2000). The average treatment time in the diet clinic was 13.3 months. The mutagenicity and the carcinogenic effects of aristolochic acid in rodent bioassays, was demonstrated two decades ago (Mengs, 1982; Mengs, 1988; Robisch *et al.*, 1982). In rats, malignant tumors were induced unusually rapidly. No HERP is reported because the human exposures were for a short time only.

### Natural pesticides

Natural pesticides, because few have been tested, are markedly underrepresented in our HERP analysis. Importantly, for each plant food listed, there are about 50 additional untested natural pesticides. Although about 10,000 natural pesticides and their break-down products occur in the human diet (Ames *et al.*, 1990a), only 72 have been tested adequately in rodent bioassays [Table 2]. Average exposures to many natural pesticides that are carcinogenic in rodents in common foods rank above or close to the median in the HERP Table, ranging up to a HERP of 0.1%. These include caffeic acid (in coffee, lettuce, tomato, apple, potato, celery, carrot, plum and pear); safrole (in spices and formerly in natural root beer before it was banned), allyl isothiocyanate (mustard), *d*-limonene (mango, orange juice, black pepper); coumarin in cinnamon; and hydroquinone, catechol, and 4-methylcatechol in coffee. Some natural pesticides in the commonly eaten mushroom (*Agaricus bisporus*) are rodent carcinogens (glutamyl-*p*-hydrazinobenzoate, *p*-hydrazinobenzoate), and the HERP based on feeding whole mushrooms to mice is 0.02%. For *d*-limonene, no human risk is anticipated because tumors are induced only in male rat kidney tubules with involvement of  $\alpha_{2u}$ -globulin nephrotoxicity, which does not appear to be relevant for humans (Hard and Whysner, 1994; International Agency for Research on Cancer, 1993; Rice *et al.*, 1999; U.S. Environmental Protection Agency, 1991c).

### Synthetic pesticides

Synthetic pesticides currently in use that are rodent carcinogens in the CPDB and that are quantitatively detected by the FDA Total Diet Study (TDS) as residues in food, are all included in Table 5. Several are at the very bottom of the ranking; however, HERP values are about at the median for 3 exposures prior to discontinuance or reduction in use: ethylene thiourea (ETU), toxaphene before its cancellation in the United States in 1982, and DDT before its ban in the United States in 1972. These 3 synthetic pesticides rank below the HERP values for many naturally occurring chemicals that are common in the diet. The HERP values in Table 5 are for residue intake by females 65 and older, since they consume higher amounts of fruits and vegetables than other adult groups, thus maximizing the exposure estimate to pesticide residues. We note that for pesticide residues in the TDS, the consumption estimates for children (mg/kg/day in 1986-1991) are within a factor of 3 of the adult consumption (mg/kg/day), greater in adults for some pesticides and greater in children for others (U.S. Food and Drug Administration, 1993b).

DDT and similar early pesticides have been a concern because of their unusual lipophilicity and persistence; however, natural pesticides can also bioaccumulate. There is no convincing epidemiological evidence of a carcinogenic hazard of DDT to humans (Key and Reeves,

1994). In a recently completed 24-year study in which DDT was fed to rhesus and cynomolgus monkeys for 11 years, DDT was not evaluated as carcinogenic (Takayama *et al.*, 1999; Thorgeirsson *et al.*, 1994) despite doses that were toxic to both liver and central nervous system. However, the protocol used few animals and dosing was discontinued after 11 years, which may have reduced the sensitivity of the study (Gold *et al.*, 1999).

Current United States exposure to DDT and its metabolites is in foods of animal origin, and the HERP value is low, 0.00008%. DDT is often viewed as the typically dangerous synthetic pesticide because it concentrates in adipose tissue and persists for years. DDT was the first synthetic pesticide; it eradicated malaria from many parts of the world, including the United States, and was effective against many vectors of disease such as mosquitoes, tsetse flies, lice, ticks and fleas. DDT prevented many millions of deaths from malaria (Jukes, 1974). It was also lethal to many crop pests, and significantly increased the supply and lowered the cost of fresh, nutritious foods, thus making them accessible to more people. DDT was also of low toxicity to humans. There is no convincing epidemiological evidence, nor is there much toxicological plausibility, that the levels of DDT normally found in the environment or in human tissues are likely to be a significant contributor to human cancer (Laden *et al.*, 2001). A recent study of breast cancer on Long Island found no association between breast cancer and blood levels of DDT, DDE, dieldrin or chlordane (Gammon *et al.*, 2002).

DDT is unusual with respect to bioconcentration, and because of its chlorine substituents it takes longer to degrade in nature than most chemicals; however, these are properties of relatively few synthetic chemicals. In addition, many thousands of chlorinated chemicals are produced in nature (Gribble, 1996). Natural pesticides can also bioconcentrate if they are fat-soluble. Potatoes, for example, naturally contain the fat soluble neurotoxins solanine and chaconine (Ames *et al.*, 1990a; Gold *et al.*, 1997b), which can be detected in the bloodstream of all potato eaters. High levels of these potato neurotoxins have been shown to cause birth defects in rodents (Ames *et al.*, 1990b).

The HERP value for ethylene thiourea (ETU), a breakdown product of certain fungicides, is the highest among the synthetic pesticide residues (0.002%), which is at the median of the ranking. The HERP value would be about 10 times lower if the potency value of the EPA were used instead of our TD<sub>50</sub>; EPA combined rodent results from more than one experiment, including one in which ETU was administered *in utero*, and obtained a weaker potency (U.S. Environmental Protection Agency, 1992a). (The CPDB does not include *in utero* exposures.) Additionally, EPA has recently discontinued some uses of fungicides for which ETU is a breakdown product, and exposure levels are therefore lower.

In 1984 the EPA banned the agricultural use of ethylene dibromide (EDB) the main fumigant in the U.S., because of the residue levels found in grain, HERP = 0.0004%. This HERP value ranks low, whereas the HERP of 140% for the high exposures to EDB that some workers received in the 1970s, is at the top of the ranking (Gold *et al.*, 1992). Two other pesticides in Table 5, toxaphene (HERP = 0.001% in 1982 and 0.0001% in 1990) and chlorobenzilate (HERP=0.0000001%), have been cancelled (Ames and Gold, 1991; U.S. Environmental Protection Agency, 1998b).

HERP values for other pesticide residues are all below the median of 0.002%. In descending order of HERP these are DDE (before the 1972 ban of DDT), ethylene dibromide, carbaryl, toxaphene (after cancellation), DDE/DDT (after the ban), dicofol, lindane, PCNB, chlorobenzilate, captan, folpet, and chlorothalonil. Some of the lowest HERP values in Table 5 are for the synthetic pesticides, captan, chlorothalonil, and folpet, which were also evaluated in 1987 by

the National Research Council (NRC) and were considered by NRC to have a human cancer risk above  $10^{-6}$  (National Research Council, 1987).

Why were the EPA risk estimates reported by NRC so high when the HERP values are so low? We have investigated this disparity in cancer risk estimation for pesticide residues in the diet by examining the two components of risk assessment: carcinogenic potency estimates from rodent bioassays and human exposure estimates (Gold *et al.*, 2001b; Gold *et al.*, 1997d). We found that potency estimates based on rodent bioassay data are similar whether calculated, as in the NRC report, as the regulatory  $q_1^*$  or as the  $TD_{50}$  in the CPDB. In contrast, estimates of dietary exposure to residues of synthetic pesticides vary enormously, depending on whether they are based on the Theoretical Maximum Residue Contribution (TMRC) calculated by the EPA vs. the average dietary residues measured by the FDA in the Total Diet Study (TDS). The EPA's TMRC is the theoretical maximum human exposure anticipated under the most severe field application conditions, which is often a large overestimate compared to the measured residues. For several pesticides, the NRC risk estimate was greater than one in a million whereas the FDA did not detect any residues in the TDS even though the TDS measures residues as low as 1 ppb (Gold *et al.*, 1997d).

In the 1980s enormous attention was given in the media to Alar, a chemical used to regulate the growth of apples while on the tree (not a pesticide). UDMH, a rodent carcinogen, is the breakdown product of Alar in apples, applesauce, and apple juice (Ames and Gold, 1989). The HERP value before use of Alar was discontinued, was 0.001%, just below the median of Table 5. Many natural dietary chemicals that are rodent carcinogens have higher HERP values, e.g. caffeic acid in lettuce, tomato, apple, and celery; safrole in spices, and catechol in coffee. Apple juice contains 353 natural volatile chemicals (Nijssen *et al.*, 1996) of which only 12 have been tested for carcinogenicity in the CPDB; 9 of these have been found to be carcinogenic.

### **Cooking and preparation of food**

Cooking and preparation of food, e.g. fermentation, also produce chemicals that are rodent carcinogens. Alcoholic beverages cause cancer in humans in the liver, esophagus and oral cavity. Epidemiological studies indicate that all types of alcoholic beverages are associated with increased cancer risk, suggesting that ethyl alcohol itself causes the effect rather than any particular type of beverage. The HERP values in Table 5 for alcohol are high in the ranking: HERP=3.6% for average U.S. consumption of all alcoholic averages combined, 1.8% in beer, and 0.6% in wine.

Cooking food is plausible as a contributor to cancer. A wide variety of chemicals are formed during cooking. Rodent carcinogens formed include furfural and similar furans, nitrosamines, polycyclic hydrocarbons, and heterocyclic amines. Furfural, a chemical formed naturally when sugars are heated, is a widespread constituent of food flavor. The HERP value for naturally-occurring furfural in average consumption of coffee is 0.006% and in white bread is 0.004%.

Recently, an industrial chemical that is also formed in cigarette smoke, was identified as a common constituent in the human diet. Acrylamide is formed when carbohydrate is cooked at high temperatures; the highest concentrations are in potato chips and French fries (Tareke *et al.*, 2002). Epidemiological studies in workers have not shown an association with cancer (Collins *et al.*, 1989; Marsh *et al.*, 1999). Acrylamide is carcinogenic at several target sites in rat bioassays, and the  $TD_{50}$  in rats is 8.89 mg/kg/day. No estimates are available for average U.S. consumption; therefore, it is not included in the HERP table (Table 5). The estimate for average con-

sumption of dietary acrylamide in Sweden is 40  $\mu\text{g/day}$  (Tareke *et al.*, 2002, <http://www.slv.se/engdefault.asp>), and the HERP value would be 0.01%. This HERP value is similar to other natural constituents of food such as safrole and furfural. Acrylamide is genotoxic, and the HERP value is above the median, suggesting that further work is needed to assess its potential hazard to humans, e.g. on formation and fate of acrylamide in food during cooking and processing, on the absorption, metabolism, and disposition in humans of acrylamide from food, on the mode of action in the animal cancer tests, and on the mechanisms of action and dose-response characteristics.

Nitrosamines are formed in food from nitrite or nitrogen oxides ( $\text{NO}_x$ ) and amines in food. Tobacco smoking and smokeless tobacco are a major source of non-occupational exposure to nitrosamines that are rodent carcinogens [*N'*-nitrosornicotine and 4-(methylnitrosamino)-1-(3-pyridyl)-1-(butanone)] (Hecht and Hoffmann, 1998). Most exposure to nitrosamines in the diet is for chemicals that are not carcinogenic in rodents (Hecht and Hoffmann, 1998; Lijinsky, 1999). The nitrosamines that are carcinogenic are potent carcinogens (Table 5), and it has been estimated that in several countries humans are exposed to about 0.3–1  $\mu\text{g}$  per day (Tricker and Preussmann, 1991) (National Academy of Sciences, 1981), primarily *N*-nitrosodimethylamine (DMN), *N*-nitrosopyrrolidine (NPYR) and *N*-nitrosopiperidine. The largest exposure was to DMN in beer: concentrations declined more than 30-fold after 1979 (HERP=0.01%) when it was reported that DMN was formed by the direct-fired drying of malt, and the industry modified the process to indirect firing (Glória, Barbour and Scanlan, 1997). By the 1990s, HERP=0.0002% (Glória *et al.*, 1997). The HERP values for average consumption of bacon are: DMN=0.0008%, *N*-Nitrosodiethylamine (DEN)=0.001%, and NPYR=0.0007%. DEN induced liver tumors in rhesus and cynomolgus monkeys and tumors of the nasal mucosa in bush babies (Thorgeirsson, *et al.*, 1994). In a study of DMN in rhesus monkeys, no tumors were induced; however the administered doses produced toxic hepatitis, and all animals died early. Thus, the test was not sensitive because the animals may not have lived long enough to develop tumors (Gold *et al.*, 1999; Thorgeirsson *et al.*, 1994).

A variety of mutagenic and carcinogenic heterocyclic amines (HA) are formed when meat, chicken or fish is cooked, particularly when charred. HA are potent mutagens with strong evidence of carcinogenicity in terms of positivity rates, multiplicity of species and target sites; however, concordance in target sites between rats and mice for these HA is generally restricted to the liver (Gold *et al.*, 1994b). Some of the target sites of HA in rats are among the more common cancer sites in humans: colon, prostate and breast. Prostate tumors were induced by PhIP at only the highest dose tested (400 ppm) and not by other HA. Under usual cooking conditions, exposures to HA are in the low ppb range, and the HERP values are low. The values in Table 5, which rank below the median, are based on hamburger consumption because hamburger has the best available concentration estimates based on various levels of doneness. A recent estimate of HA in the total diet was about 2-fold higher than our consumption estimates for hamburger (Bogen and Keating, 2001; Keating and Bogen, 2001).

For HA in pan fried hamburger, the HERP value is highest for PhIP, 0.0002%, compared to 0.00003% for MeIQx and 0.00001% for IQ. Carcinogenicity of the three HA in the HERP table, IQ, MeIQx, and PhIP, has been investigated in studies in cynomolgus monkeys. IQ rapidly induced a high incidence of hepatocellular carcinoma (Adamson *et al.*, 1994) and the HERP value would be 2.5 times higher than rats. MeIQx, which induced tumors at multiple sites in rats and mice (Gold *et al.*, 1997c), did not induce tumors in monkeys (Ogawa *et al.*, 1999). The PhIP study is still in progress. Metabolism studies indicate the importance of *N*-

hydroxylation in the carcinogenic effect of HA in monkeys (Ogawa *et al.*, 1999; Snyderwine *et al.*, 1997).

### Food Additives

Food additives that are rodent carcinogens can be either naturally-occurring (e.g., allyl isothiocyanate, furfural) or synthetic (e.g., butylated hydroxyanisole [BHA] and saccharin). The highest HERP values for average dietary exposures to synthetic rodent carcinogens in Table 5 are for exposures in the early 1970s to BHA (0.01%) and saccharin in the 1970s (0.005%). Both are nongenotoxic rodent carcinogens for which data on mechanism of carcinogenesis strongly suggest that there would be no risk to humans at the levels found in food (See *Saccharin* below).

### Naturally-occurring food additives

For five naturally-occurring rodent carcinogens that are also produced commercially and used as food additives, average exposure data were available and they are included in Table 5. The HERP value for the natural occurrence of each chemical is greater than for use as a commercial additive because the natural exposures are greater. For furfural (a product of cooking discussed above) the HERP value for the natural occurrence is 0.03% compared to 0.0003% for the additive; for *d*-limonene the natural occurrence (e.g. in citrus and other common foods) HERP is 0.1% compared to 0.007% for the additive; for estragole (in spices) the natural occurrence HERP is 0.001% compared to 0.0002% for the additive; for methyleugenol the natural occurrence (in spices) HERP is 0.004% compared to 0.0006% for the additive. For allyl isothiocyanate the natural occurrence HERP in mustard is 0.0003% compared to 0.0002% for the additive; the natural value only includes mustard (Krul *et al.*, 2002; Tsao *et al.*, 2002), but allyl isothiocyanate is also present in other *Brassica* vegetables, e.g., cabbage, cauliflower, and Brussels sprouts (Nijssen *et al.*, 1996).

Safrole is the principle component (up to 90%) of oil of sassafras. It was formerly used as the main flavor ingredient in root beer. It is also present in the oils of basil, nutmeg, and mace (Nijssen *et al.*, 1996). The HERP value for average consumption of naturally-occurring safrole in spices is 0.03%. Safrole and safrole-containing sassafras oils were banned from use as food additives in the U.S. and Canada (Canada Gazette, 1995; U.S. Food and Drug Administration, 1960). For a person consuming a glass of sassafras root beer per day for life (before the 1964 ban in the U.S.), the HERP value would have been 0.2% (Ames *et al.*, 1987). Sassafras root can still be purchased in health food stores and can therefore be used to make tea; the recipe is on the World Wide Web.

### Butylated hydroxyanisole (BHA)

BHA is a phenolic antioxidant that is Generally Regarded as Safe (GRAS) by the FDA. By 1987, after BHA was shown to be a rodent carcinogen, its use declined six fold (HERP=0.002%) (U.S. Food and Drug Administration, 1991a); this was due to voluntary replacement by other antioxidants, and to the fact that the use of animal fats and oils, in which BHA is primarily used as an antioxidant, has consistently declined in the U.S. The mechanistic and carcinogenicity results on BHA indicate that malignant tumors were induced only at a dose above the MTD at which cell division was increased in the forestomach, which is the only site of tumorigenesis; the proliferation is only at high doses, and is dependent on continuous dosing until late in the experiment (Clayson *et al.*, 1990). Humans do not have a forestomach. We note that the dose-response for BHA curves sharply upward, but the potency value used in HERP is based on a linear model; if the California EPA potency value (which is based on a linearized multistage model)



were used in HERP instead of  $TD_{50}$ , the HERP values for BHA would be 25 times lower (California Environmental Protection Agency. Standards and Criteria Work Group, 1994). A recent epidemiological study in the Netherlands found no association between BHA consumption and stomach cancer in humans (Botterweck *et al.*, 2000).

### **Saccharin**

Saccharin, which has largely been replaced by other sweeteners, has been shown to induce tumors in rodents by a mechanism that is not relevant to humans. Recently, both NTP and IARC re-evaluated the potential carcinogenic risk of saccharin to humans. NTP delisted saccharin in its *Report on Carcinogens* (U.S. National Toxicology Program, 2000b), and IARC downgraded its evaluation to Group 3, “not classifiable as to carcinogenicity to humans” (International Agency for Research on Cancer, 1999b). There is convincing evidence that the induction of bladder tumors in rats by sodium saccharin requires a high dose and is related to development of a calcium phosphate-containing precipitate in the urine (Cohen, 1995), which is not relevant to human dietary exposures. In a 24-year study by NCI, rhesus and cynomolgus monkeys were fed a dose of sodium saccharin that was equivalent to 5 cans of diet soda daily for 11 years (Thorgeirsson *et al.*, 1994). The average daily dose-rate of sodium saccharin was about 100 times lower than the dose that was carcinogenic to rats (Gold *et al.*, 1999; Gold *et al.*, 1997c). There was no carcinogenic effect in monkeys. There was also no effect on the urine or urothelium, no evidence of increased urothelial-cell proliferation or of formation of solid material in the urine (Takayama *et al.*, 1998). One would not expect to find a carcinogenic effect under the conditions of the monkey study because of the low dose administered (Gold *et al.*, 1999). Additionally, however, there may be a true species difference because primate urine has a low concentration of protein and is less concentrated (lower osmolality) than rat urine (Takayama *et al.*, 1998). Human urine is similar to monkey urine in this respect (Cohen, 1995).

### **Mycotoxins**

Of the 23 fungal toxins tested for carcinogenicity, 14 are positive (61%) (Table 4). The mutagenic mold toxin, aflatoxin, which is found in moldy peanut and corn products, interacts with chronic hepatitis infection in human liver cancer development (Qian *et al.*, 1994). There is a synergistic effect in the human liver between aflatoxin (genotoxic effect) and the hepatitis B virus (cell division effect) in the induction of liver cancer (Wu-Williams *et al.*, 1992). The HERP value for aflatoxin of 0.008% is based on the rodent potency. If the lower human potency value calculated by FDA from epidemiological data were used instead, the HERP would be about 10-fold lower (U.S. Food and Drug Administration, 1993a). Aflatoxin also induced liver tumors in cynomolgus and rhesus monkeys, and the HERP value using  $TD_{50}$  in monkeys would be between the value for rodents and humans. Biomarker measurements of aflatoxin in populations in Africa and China, which have high rates of hepatitis B and C viruses and liver cancer, confirm that those populations are chronically exposed to high levels of aflatoxin (Groopman *et al.*, 1992; Pons, 1979). Liver cancer is unusual in the U.S. and Canada (about 2% of cancer deaths), and is more common among men than women (National Cancer Institute of Canada, 2001; Ries *et al.*, 2000). In the U.S. an increase in liver cancer in the early 1990s was most likely due to the spread of hepatitis virus infection transmitted by transfusions (before screening of blood products for HCV) use of intravenous drugs, and sexual practices ten to 30 years earlier (El-Serag and Mason, 1999; Ince and Wands, 1999). In the U.S., one study estimated that hepatitis viruses can account for half of liver cancer cases among non-Asians and even more among Asians (Yu *et al.*, 1991).

Ochratoxin A, a potent rodent carcinogen (Gold and Zeiger, 1997), has been measured in Europe and Canada in agricultural and meat products. An estimated exposure of 1 ng/kg/day would have a HERP value at about the median of Table 5 (International Life Sciences Institute, February 1996; Kuiper-Goodman and Scott, 1989).

### **The Persistent Contaminants PCBs and TCDD**

Polychlorinated biphenyls (PCBs) and tetrachlorodibenzo-*p*-dioxin (TCDD, dioxin), which have been a concern because of their environmental persistence and carcinogenic potency in rodents, are primarily consumed in foods of animal origin. In the U.S. PCBs are no longer used, but some exposure persists. Consumption in food in the U.S. declined about 20-fold between 1978-1986 (Gartrell *et al.*, 1986; Gunderson, 1995). PCBs, which are not flammable, were formerly used as coolants and lubricants in electrical equipment. The HERP value for PCB in Table 5 for the most recent reporting in the FDA Total Diet Study (1984-86) is 0.00008%, towards the bottom of the ranking, and far below many values for naturally occurring chemicals in common foods. It has been reported that some countries may have higher intakes of PCBs than the U.S. (World Health Organization, 1993). A recent epidemiological study found no association between PCBs and breast cancer, in which PCBs were measured in the blood of women on Long Island (Gammon *et al.*, 2002).

TCDD, the most potent rodent carcinogen, is produced naturally by burning when chloride ion is present, e.g. in forest fires or wood burning in homes. EPA (U.S. Environmental Protection Agency, 2000) estimates that the source of TCDD is primarily from the atmosphere directly from emissions, e.g. incinerators or burning trash, or indirectly by returning dioxin that is already in the environment to the atmosphere (U.S. Environmental Protection Agency, 1994a; U.S. Environmental Protection Agency, 2001). TCDD bioaccumulates through the food chain because of its lipophilicity, and more than 95% of human intake is from animal fats in the diet (U.S. Environmental Protection Agency, 2001). Dioxin emissions decreased by 75% from 1987-1995, which EPA primarily attributes to reduced medical and municipal incineration emissions. The decline continues (U.S. Environmental Protection Agency, 2001). Estimates of dietary intake can vary because TCDD is often not detected in samples of animal products (about 60% of such samples have no detectable TCDD). Intake estimates are based on an assumption that dioxin is present in food at one-half the limit of detection when no dioxin is detected; the intake estimate would be lower by about half if zero were assumed instead (Schecter *et al.*, 2001).

TCDD, which is not genotoxic (U.S. Environmental Protection Agency, 2000), exerts many of its harmful effects in experimental animals through binding to the Ah receptor (AhR), and does not have effects in the AhR knockout mouse (Birnbaum, 1994; Fernandez-Salguero *et al.*, 1996). A wide variety of natural substances also bind to the Ah receptor (e.g., tryptophan oxidation products), and insofar as they have been examined, they have similar properties to TCDD (Ames *et al.*, 1990b) including inhibition of estrogen-induced effects in rodents (Safe *et al.*, 1998). For example, a variety of flavones and other plant substances in the diet and their metabolites bind to the receptor or are converted in the stomach to chemicals that bind to the Ah receptor e.g. indole-3-carbinol (I3C). I3C is the main metabolite of glucobrassicin, a natural chemical that is present in large amounts in vegetables of the *Brassica* genus, including broccoli, and gives rise to the potent Ah binder, indole carbazole (Bradfield and Bjeldanes, 1987). In comparing possible harmful effects, the binding affinity (greater for TCDD) and amounts in the diet (much greater for dietary compounds) both need to be considered. Some studies provide evidence that I3C enhances carcinogenicity (Dashwood, 1998). Additionally, both I3C and

TCDD, when administered to pregnant rats, resulted in reproductive abnormalities in male offspring (Wilker *et al.*, 1996). Currently, I3C is in clinical trials for prevention of breast cancer (Kelloff *et al.*, 1996a; Kelloff *et al.*, 1996b; U.S. National Toxicology Program, 2000a) and is also being tested for carcinogenicity by NTP (U.S. National Toxicology Program, 2000a). I3C is marketed as a dietary supplement at recommended doses about 30 times higher (Therapeutics, 2000) than present in the average Western diet (U.S. National Toxicology Program, 2000a).

TCDD has received enormous scientific and regulatory attention, and controversy abounds about possible health risks to humans. It has been speculated that nearly 7000 publications have been written and \$3-5 billion US has been spent to assess dioxin exposure and health effects to humans and wildlife (Paustenbach, 2002). The U.S. EPA has been estimating dioxin cancer risk since 1991 (U.S. Environmental Protection Agency, 1994a; U.S. Environmental Protection Agency, 1994b; U.S. Environmental Protection Agency, 1995; U.S. Environmental Protection Agency, 2000), and the EPA Science Advisory Board has recently recommended re-consideration of many issues in the EPA assessment. (Paustenbach, 2002; Science Advisory Board, 2001). A committee of the U.S. National Academy of Sciences has been appointed to evaluate the risks from dioxins in the diet.

The IARC evaluated TCDD as a human carcinogen (Group 1) on the basis of overall cancer mortality, even though no specific type of cancer was found to be increased in the epidemiological studies of formerly highly-exposed workers (International Agency for Research on Cancer, 1997). An IARC evaluation based on overall cancer mortality is unprecedented. With respect to risks, IARC concluded that “Evaluation of the relationship between the magnitude of the exposure in experimental systems and the magnitude of the response (i.e. dose-response relationships) do not permit conclusions to be drawn on the human health risks from background exposures to 2,3,7,8-TCDD.” (International Agency for Research on Cancer, 1997, p. 342) The U.S. NTP *Ninth Report on Carcinogens* concurred with IARC in the human carcinogen evaluation (U.S. National Toxicology Program, 2000b; U.S. National Toxicology Program, 2001). The EPA characterized TCDD as a “human carcinogen” but concluded that “there is no clear indication of increased disease in the general population attributable to dioxin-like compounds.” (U.S. Environmental Protection Agency, 2000; U.S. Environmental Protection Agency, 2001) One meta-analysis combined the worker studies and found that there was no increasing cancer mortality, overall or for a specific organ, with increasing exposure to TCDD (Starr, 2001). The most recent meta-analysis, using additional followup data, found an increased trend in total cancer mortality with increasing TCDD exposure (Crump *et al.*, 2003).

Worldwide, dioxin has primarily been regulated by many groups on the basis of sensitive reproductive and developmental (noncancer) effects in experimental animals, which have a threshold. In contrast the U.S. EPA estimates have used cancer potency factors and a standard linear risk assessment model. The level of acceptable intake for humans has been judged similarly by many groups: the World Health Organization (Van den Berg *et al.*, 1998), the U.S. Agency for Toxic Substances and Disease Registry (ATSDR) (Agency for Toxic Substances and Disease Registry, 1998), the European Community (European Commission Scientific Committee on Foods, 2001), Health and Welfare Canada (Ministry of Environment and Energy, 1997), and the Japanese Environmental Agency (Japanese Environmental Agency, 1999). The acceptable level set by these groups differs from the U.S. EPA assessments that are based on cancer: the risks levels that are considered to be safe are 1,000 to 10,000 times higher (less stringent) than the levels that the EPA draft documents would consider to be a negligible risks (one in a million cancer risk). All of the agencies, including the U.S. EPA, have based their evaluations on Toxic

Equivalency (TEQ), a method which combines exposures to all dioxins and dioxin-like compounds. These agencies also take into consideration the body burden doses of dioxins in humans due to bioaccumulation in lipid. There are uncertainties in these methods, e.g., the TEQ method assumes that the toxic effects of many compounds are additive; however antagonistic effects have been reported among these chemicals in experimental studies (European Commission Scientific Committee on Foods, 2000). The EPA risk estimates thus provide a worst-case risk; actual risks are unlikely to be greater and may be substantially less. The EPA Science Advisory Board (SAB) has recommended reconsideration of many aspects of the EPA cancer risk assessment, including the classification as a known human carcinogen, methods to estimate cancer potency and noncancer effects, uncertainties in estimation of body burden of dioxins, and consideration of dose-response curves other than a linear one (Agency for Toxic Substances and Disease Registry, 1998; Paustenbach, 2002; Science Advisory Board, 2001).

In table 5, the HERP value of 0.0003%, which is for average U.S. intake of TCDD, is below the median of the values in table 5. If the exposures to all dioxin-like compounds were used for the exposure estimate (TEQ), then the HERP value would be 10 times greater. If the body burden of these combined dioxins were also considered in HERP as the EPA has done, then the combined effect of these two factors would make the HERP value 30 times greater, (HERP would be 0.01%, but would not be comparable to the other HERP values in table 5 because of combining exposures to several chemicals [TEQ] and considering exposure due to bioaccumulation).

In sum, the HERP analysis in Table 5 demonstrates the ubiquitous exposures to rodent carcinogens in everyday life, and documents that possible hazards to the background of naturally-occurring rodent carcinogens occur throughout the ranking. Widespread exposures to naturally-occurring rodent carcinogens cast doubt on the relevance to human cancer of low-level exposures to synthetic rodent carcinogens. In regulatory efforts to prevent human cancer, the evaluation of low-level exposures to synthetic chemicals has had a high priority. Our results indicate, however, that a high percentage of both natural and synthetic chemicals are rodent carcinogens at the MTD, that tumor incidence data from rodent bioassays are not adequate to assess low-dose risk. Moreover, there is an imbalance in testing of synthetic chemicals compared to natural chemicals. There is a background of natural chemicals in the diet that rank at or near the median HERP value, even though so few natural chemicals have been tested in rodent bioassays. In Table 5, 90% of the HERP values are above the level that has been used for as the virtually safe dose (VSD) in regulatory policy for rodent carcinogens.

Caution is necessary in drawing conclusions from the occurrence in the diet of natural chemicals that are rodent carcinogens. It is not argued here that these dietary exposures are necessarily of much relevance to human cancer. The major known causes of human cancer are not single chemicals agents like those studied in rodent bioassays (Misconception 2).

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### Appendix: Calculation of HERP

The HERP index takes into account both human exposures and the carcinogenic dose to rodents and compares them. HERP values indicate what percentage of the rodent carcinogenic daily dose (mg/kg/day) for 50% of test animals, a person receives from an average daily exposure (mg/kg/day).

For example, methyleugenol is a chemical that is carcinogenic in rats and mice and has a HERP value of 0.004% for average daily U.S. exposure in food from its natural occurrence, and 0.0006% for average daily U.S. exposure as a synthetic food additive. Below is an example of the HERP calculation for methyleugenol that occurs naturally (see Table 6 at HERP=0.004%). Data are available indicating that average naturally occurring methyleugenol consumption in the U.S. is 46.2  $\mu\text{g}/\text{day}$  (Smith *et al.*, 2002). The calculation of HERP from the values in Table 6 for methyleugenol is as follows:

1. Human dose of rodent carcinogen is  $46.2 \mu\text{g}/\text{day} / 70 \text{ kg body weight} = 0.66 \mu\text{g}/\text{kg}/\text{day}$  (=0.00066 mg/kg/day).
2. Rodent potency: the  $\text{TD}_{50}$  is 18.6 mg/kg/day in mice
3. Possible hazard (HERP) is:  $\frac{0.0006 \text{ mg/kg/day human exposure}}{18.6 \text{ mg/kg/day } \text{TD}_{50}} = 0.00004$ ;

$$0.00004 \times 100 = 0.004\%$$

The  $\text{TD}_{50}$  values used in HERP are averages for rats and mice separately, calculated by taking the harmonic mean of the  $\text{TD}_{50}$  values from positive experiments. For methyleugenol the  $\text{TD}_{50}$  in rats is 19.7 mg/kg/day and in mice 18.6 mg/kg/day. Since the mouse  $\text{TD}_{50}$  is lower (more potent), this value is used in HERP. Experiments in the CPDB that do not show an increase in tumors are ignored in HERP.

The  $\text{TD}_{50}$  value for rats or mice in the HERP table is a harmonic mean of the most potent  $\text{TD}_{50}$  values from each positive experiment. The harmonic mean ( $T_H$ ) is defined as:

$$T_H = \frac{1}{\frac{1}{n} \sum_{i=1}^n \frac{1}{T_i}}$$

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